SYNTHESIS OF CERTAIN 2-AROYLINDOLE DERIVATIVES OF POTENTIAL ANALGESIC, ANTI-INFLAMMATORY AND ANTIPYRETIC ACTIVITIES

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تتناول هذه الدراسة تخليق بعض مشتقات - أرويل إندول والتأكد من صحة التراكيب البنائية المركبات الجديدة بواسطة التحليل الدقي للعناصر وبعض القياسات الطيفية مثل الأشعة تحت الحمراء وكذلك الرنين النووي المغناطيسي وبعض منها باشعة مطياف الكتلة أما فيما يختص بالفاعلا وكذلك الرنين النووي المغناطيسي وبعض منها باشعة مطياف الكتلة أما فيما يختص بالفاعلا ولفار ماكولوجية المبدئية فقد تمت دراسة تأثير كافة مشتقات الإندول المحضرة (Ga-g, 7a-f, 8a-f, 10a-13a) الفار ماكولوجية المبدئية فقد تمت دراسة تأثير كافة مشتقات الإندول المحضرة بالجارب. أظهرت (Ba-g, 7a-f, 8a-f, 10a-13a) وكذلك الرنين النووي المغناطيسي وبعض منها باشعة مطياف الكتلة أما فيما يختص بالفاعلا الفار ماكولوجية المبدئية فقد تمت دراسة تأثير كافة مشتقات الإندول المحضرة بالجارب. أظهرت (Ga-g, 7a-f, 8a-f, 10a-13a) وكندي الاستبدال علي نوية الإندول بمجموعة فينيل علي الموضع ومجموعة (بارا هالو) نزويل علي الموضع ما ومجموعة (بارا هالو) نزويل علي الموضع ما ومحموعة (بارا هالو) نزويل علي الموضع المركبات لتأثيرها كمسكنات للألم وأظهر المركبان (Ab وأظهر النتائج في هذه المركبات لتأثيرها كمسكنات للألم وأظهر المركبين الموضع الموضع المركبين فالموضع المركبين فالموضع الموريان في هذه المركبات لتأثيرها كمسكنات للألم وأظهر المركبين الموضع الموضع الموريان في هذه المركبين الموضع الموريان فاعلية ألمركبين المركبين فالمين والموريان فالمين المركبين (Ab والموريان فالمين المركبين (Ab وألهر المركبين فالموريان في هذه المركبين في ألموريان التجارب أطهرت التائم أن كلا المركبين (Ab وألهر الخوري التهاب بصورة أسرع وأكثر فاعلية ولمدة أطول مقارنة والمركبين في قاما بتخفيض الحرارة وتقليل الالتهاب بصورة أسرع وأكثر فاعلية ولمدة أطول مقارنة التهاب المركبين قد قاما بتخفيض الحرارة وتقليل الالتهاب بصورة أسرع وأكثر فاعلية ولمدة أطول مقارنة بعقار الأريدوميثازين. كذلك تم اختبار تأثير كلا المركبين (Ab وألموريا ألمرين التهارين التجارب كمسببين المركبين قد المركبين ما حقن الخميرة في جرذان التجارب ألمون المركبين فالموريا المودة ألمور المركبين كلا المركبين فالمولية ألمول المركبين كالمركبين في ألمول المرادة ألمول المركبين المركبين ألمول مقارنة بعقار الإندوميثازين عندكل المرعات المستخدمة.

In the present study, a series of 2-aroylindole derivatives were synthesized by phase transfer catalysis (PTC) and were characterized by IR, ¹H-NMR, Mass spectral and Elemental analysis. Indole derivatives **6a-g**, **7a-f**, **8a-f** and 9a-13a were tested for analgesic activity using hot-plate test. Compounds 7b and 8b were tested for antipyretic and anti-inflammatory activity using yeast induced hyperthermia and paw edema in rats. Analgesic activity was shown when indole nucleus was substituted at position 2 and 3 by phenyl and (p-halo)benzoyl moieties respectively, where highest activity was recognised in compounds 7b and 8b. Both compounds also exhibited faster, more effective and prolonged reduction in hyperthermia and edema induced in rats compared with indomethacin. Compounds **7b** and **8b** were also tested for ulcerogenic activity in mice, where a lower ulcerogenic effect was observed compared with indomethacin at all tested dose levels.

INTRODUCTION

Synthesis, reactivity and wide range of biological activities found among indoles have been a topic of research interest. Great interest has been focused on indomethacin as a drug used in the treatment of rheumatoid arthritis and related conditions. Modification of its structure, specifically, at nitrogen or variations of attachment position of the benzoyl moiety at position 2 or 3 of the indole nucleus led to antiinflammatory and analgesic compounds.¹⁻¹⁰ In addition, the anti-inflammatory potencies of zomepirac, tolmetine and ketorolac are well-known. This class of compounds is characterized by an aryl-heteroaryl ketone arrangement.¹¹⁻¹⁴

Based on the above-mentioned findings, it seemed reasonable to synthesize a series of 2aroylindoles that incorporate the methanone moiety linking the aryl and the heteroaryl (indole) nucleus. Selection of the different

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substituents was based on the following rationale:

- a) Simple alkyl groups of increasing carbon chain (e.g. methyl, ethyl, n-propyl, n-butyl) would find out whether *N*-alkylation and/or chain lengthening has any effect on biological response.
- b) The *p*-substituent (H, Cl, Br, CH_3 , NO_2) of the benzoyl entity would evaluate the *p*-substituent effect. These substituents were capable to correlate the physico-chemical parameters with the anticipated biological activity, if any.
- c) The *N*-benzyl moiety would explore the effect of introducing an aryl entity separated by one carbon from the nitrogen atom of the indole ring.
- d) The *N*-benzoyl group was chosen to be of closer structural approach to indomethacin.
- e) Substitution at position 3 of the indole nucleus by methyl or phenyl was chosen to

explore the importance of existence of arylaroyl arrangement for biological response.

Chemistry

Synthesis of 1-alkyl-2-aroylindoles (**6b-f**, **7b-f**, **8b-f** and **9b-f**) was attempted by two different approaches. The first one was through cyclo-condensation of *N*-alkyl-*o*-aminobenzo-phenone (**1b,c**) with appropriate phenacyl bromide (**5a,d**), leading to compounds **6c** and **9b**, respectively, Fig 1. The second approach involved *N*-alkylation of the already cyclized 2-aroylindoles (**6a-13a**) using solid-liquid phase transfer catalysis (SL-PTC); resulting in compounds {**6-9** (**b-f**)}, Fig. 1.

2-Aroylindoles (**6a-13a**) in their turn were obtained by treating the appropriate *o*aminophenyl ketones with suitable phenacyl bromide, Fig.1. Moreover, synthesis of 1,2-Dibenzoyl-3-phenylindole (**6g**) was achieved through base-catalyzed acylation of 2-benzoyl-3-phenylindole (**6a**), Fig. 1.



Fig. 1: Synthesis of 1-alkyl-2-aroylindoles (first approach) & N-alkylation and N-acylation of 2-aroylindoles (second approach).

Pharmacology

The effect of synthesized compounds on mean licking time in hot-plate test was used as measure of analgesic activity. Also, antipyretic activity of compounds 7b and 8b was investigated using veast induced hyperthermia test. In addition, both the time course of edema and percentage inhibition of edema were used as measures for antiinflammatory activity of compounds 7b and 8b using yeast induced edema. Activities of the tested compounds were compared with indomethacin as a reference drug at 5 and 10 mg/kg dose levels. Compounds 7b and 8b were then tested for ulcerogenic activity in mice.

EXPERIMENTAL

Chemistry

M.p were determined in open capillary tubes and are uncorrected. Precoated silica gel plates (kieselgel 60 GF, Merck, Germany) were used for TLC. IR spectra were recorded (in KBr) on a Shimadzu 200 spectrophotometer. ¹H-NMR spectra were recorded on Varian EM-360 60 MHz NMR Spectrometer using CDCl₃ as a solvent and were reported as δ values in ppm relative to tetramethylsilane (TMS) as internal standard. Elemental analysis was performed on Elementar Analysenysteme GmbH VarioEL V2.3 07. Analyses indicated by the symbols of the elements are within $\pm 0.4\%$ of the theoretical values. Mass spectra were performed on a JOEL JMS600 spectrometer.

General procedure for N-tosyl- oaminophenyl ketones 3a and 4a¹⁵

Tosyl chloride (0.1 mol) dissolved in chloroform (50 ml) was added dropwise to a stirred solution of *o*-aminophenyl ketone (**1a**, **2a**; 0.1 mol) and pyridine (0.1 mol) in chloroform (250 ml) at 0°. The reaction mixture was stirred for 6 hrs then washed with water (2x40 ml), HCl (2x30 ml, 10%) and water (2x40 ml). The organic layer was separated, dried using anhydrous sodium sulfate and evaporated under vacuum leaving a solid which was crystallized from ethanol.

N-Tosyl-o-aminobenzophenone (**3a**) Yield 85%, m.p 127-129°. IR (KBr, cm⁻¹): 3250 (N-H), 1644 (C=O), 1394, 1184 (SO₂). ¹H-NMR (CDCl₃, , ppm): 2.2 (s, 3H, tolyl CH₃), 6.9-7.9

(m, 13H, tolyl C_6H_4 , C_6H_5 and *o*-disubstituted C_6H_4), 10.0 (s, 1H, NH; D₂Oexchange). Anal. Calc. for $C_{20}H_{17}NO_3S$: C, 68.36; H, 4.88; N, 3.99; S, 9.12. Found: C, 68.34; H, 4.66; N, 3.96; S, 9.0%.

N-Tosyl-o-aminoacetophenone (**4a**) Yield: 82%, m.p 154-157 °C. IR (KBr, cm⁻¹): 3350 (N-H), 1646 (C=O), 1338, 1164 (SO₂). ¹H-NMR (CDCl₃, , ppm): 2.4 (s, 3H, tolyl CH₃), 2.5 (s, 3H, COCH₃), 7.2-7.5 (m, 8H, tolyl C₆H₄ and odisubstituted C₆H₄), 11.5 (s, 1H, NH; D₂Oexchange). Anal. Calc. for C₁₅H₁₅NO₃S: C, 62.26; H, 5.23; N, 4.84; S, 11.08. Found: C, 62.17; H, 5.38; N, 4.86; S, 10.81%.

General procedure for N-alkyl-N-tosyl-oaminophenyl ketones 3 and 4 (b,c)¹⁵

The appropriate alkyl halide (0.02 mol) was added to a suspension of N-tosyl-oaminophenyl ketone 3a, 4a (0.01 mol) and potassium carbonate (0.02 mol) in acetone (30 ml). The reaction was stirred for 18-20 hrs at r.t. The suspension was evaporated under vacuum and the residue was extracted with chloroform (5x30 ml). The combined chloroform extracts was washed with water (3x20 ml), dried with anhydrous sodium sulfate and evaporated under vacuum. The residue was crystallized from ethanol.

N-Methyl-N-tosyl-o-aminobenzophenone (**3b**) Yield: 84%, m.p 126-128°. IR(KBr, cm⁻¹): 1665 (C=O), 1337, 1157 (SO₂). ¹H-NMR (CDCl₃, , ppm): 2.4 (s, 3H, tolyl CH₃), 3.1 (s, 3H, NCH₃), 6.8-7.9 (m, 13H, tolyl C₆H₄, C₆H₅ and *o*-disubstituted C₆H₄). Anal. Calc. for C₂₁H₁₉NO₃S: C, 69.02; H, 5.24; N, 3.83; S, 8.77. Found: C, 68.63; H, 5.16; N, 3.79; S, 8.58%.

N-Methyl-N-tosyl-o-aminoacetophenone (**4b**) Yield: 82%, m.p 116-118°. IR (KBr, cm⁻¹): 1685 (C=O), 1339, 1170. (SO₂). ¹H-NMR (CDCl₃, , ppm): 2.4 (s, 3H, tolyl CH₃), 2.7 (s, 3H, COCH₃), 3.3 (s, 3H, NCH₃), 6.7-7.4 (m, 8H, tolyl C₆H₄ and *o*-disubstituted C₆H₄). Anal. Calc. for C₁₆H₁₇NO₃S: C, 63.34; H, 5.65; N, 4.62; S, 10.57. Found: C, 63.41; H, 5.39; N, 4.61; S, 10.45%.

N-Ethyl-N-tosyl-o-aminobenzophenone (**3c**) Yield: 91%, m.p 102-104°. IR (KBr, cm⁻¹): 1662 (C=O), 1339, 1169 (SO₂). ¹H-NMR (CDCl₃, , ppm): 1.2 (t, 3H, NCH₂CH₃), 2.3 (s, 3H, tolyl CH₃), 3.7 (q, 2H, NCH₂CH₃), 7-7.9 (m, 13H, tolyl C₆H₄, C₆H₅ and *o*-disubstituted C_6H_4). Anal. Calc. for $C_{22}H_{21}NO_3S$: C, 69.63; H, 5.58; N, 3.69; S, 8.45. Found: C, 69.42; H, 6.00; N, 3.66; S, 8.33%.

N-Ethyl-N-tosyl-o-aminoacetophenone

(4c) Yield: 96%, m.p 136-138 °C. IR (KBr, cm ¹): 1696 (C=O), 1338, 1169 (SO₂). ¹H-NMR (CDCl₃, , ppm): 1.2 (t, 3H, NCH₂CH₃), 2.5 (s, 3H, tolyl CH₃), 2.7 (s, 3H, COCH₃), 3.7 (q, 2H, NCH₂CH₃), 6.8-7.5 (m, 8H, tolyl C₆H₄ and *o*-disubstituted C₆H₄). Anal. Calc. for C₁₇H₁₉NO₃S: C, 64.33; H, 6.03; N, 4.41; S, 10.10. Found: C, 64.23; H, 5.78; N, 4.41; S, 10.20%.

Detosylation reaction- General procedure for preparation of N- alkyl- o-aminophenyl ketones 1 and 2 (b,c)

To N-alkyl-N-tosyl-*o*-aminophenyl ketone **3** and **4** (**b**, **c**); (0.01 mol) was added sulfuric acid (5 ml, 80%). The mixture was heated with stirring at 155-65° and stirred for about 5 min until the solid passed into solution. The solution was left to cool and poured onto water (25-30 ml). The solution was rendered alkaline using a solution of sodium hydroxide (20%) and left overnight in refrigerator. The residue obtained was either crystallized from ethanol (**1b**, **c**) or separated using dry column flash chromatography (**2b**, **c**) where the elution system was hexane: ethyl acetate (9:1).

N-Methyl-o-aminobenzoophenone (1b) Yield: 80%, m.p 70-72°. IR (KBr, cm⁻¹): 3355 (N-H), 1618 (C=O), 740, 697 (bending of mono substituted Ar-H). ¹H-NMR (CDCl₃, , ppm): 3.0 (d, 3H, NCH₃), 6.5-7.8 (m, 9H, C₆H₅ and *o*-disubstituted C₆H₄), 8.6 (hump, 1H, NH; D₂Oexchange). Anal. Calc. for C₁₄H₁₃NO: C, 79.59; H, 6.20; N, 6.63. Found: C, 78.95; H, 6.26; N, 6.54%.

N-Methyl-o-aminoacetophenone (2b) Yield: 82%, oil. IR (KBr, cm⁻¹): 3315 (N-H), 1634 (C=O). ¹H-NMR (CDCl₃, , ppm): 2.6 (s, 3H, COCH₃), 2.9 (d, 3H, NCH₃), 6.7-7.7 (m, 4H, *o*-disubstituted C_6H_4), 8.8 (hump, 1H, NH; D₂Oexchange).

N-Ethyl-o-aminobenzoophenone (1c) Yield: 81%, m.p 43-45°. IR (KBr, cm⁻¹): 3335 (N-H), 1617 (C=O), 750, 702 (Ar-H). ¹H-NMR (CDCl₃, , ppm): 1.3 (t, 3H, NCH₂CH₃), 3.2 (m, 2H, NCH₂CH₃), 6.3-6.8; 7.2-7.7 (m, 9H, C_6H_5 and *o*-disubstituted C_6H_4), 8.5 (hump, 1H, NH; D₂Oexchange). N-Ethyl-o-aminoacetophenone (**2c**) Yield: 82%, oil. IR (KBr, cm⁻¹): 3315 (N-H), 1633 (C=O). ¹H-NMR (CDCl₃, , ppm): 1.3 (t, 3H, NCH₂CH₃), 2.6 (s, 3H, COCH₃), 3.2 (m, 2H, NCH₂CH₃), 6.7-7.5; 7.2-7.7 (m, 4H, *o*disubstituted C_6H_4), 8.8 (hump, 1H, NH; D₂Oexchange).

General procedure for preparation of 2substituted benzoyl-3- methyl/ phenylindoles 6-13 (a)

The appropriate *o*-aminophenylketone (1, 2 (a), 0.02 mol) and the appropriate phenacyl bromide (**5a-e**, 0.02 mol) were dissolved in anhydrous DMF (50 ml). The solution was heated in an oil bath (80-90°) for 12-16 hrs, monitored by TLC [hexane: ethyl acetate (9:1)]. The reaction mixture was then poured on ice-water. The residue was filtered, dried and crystallized from methanol.

2-Benzoyl-3-phenyl-1H-indole (**6a**) Yield: 73%, m.p $203-204^{\circ}$ as reported.¹⁶

2-(4-Bromobenzoyl)-3-phenyl-1H-indole (7a) Yield: 60%, m.p 188-189°. IR (KBr, cm⁻¹): 3330 (N-H), 1610 (C=O), 763, 712 (Ar-H). ¹H-NMR (CDCl₃, , ppm): 7.1-7.9 (m, 13H, C₆H₅, COC₆H₄Br, *o*-disubstituted C₆H₄), 9.5 (hump, 1H, NH). Anal. Calc. for C₂₁H₁₄BrNO: C, 67.04; H, 3.75; N, 3..72. Found: C, 67.04; H, 3.34; N, 3.72%. MS (FAB+), m/z: 376.98 (0.5%, M+H), 378.92 (0.3%, M+H+2).

2-(4-Chlorobenzoyl)-3-phenyl-1H-indole (**8a**) Yield: 55%, m.p 179-180° as reported.¹⁷

2-(4-Methylbenzoyl)-3-phenyl-1H-indole (**9a**) Yield: 83%, m.p 156-158° (reported m.p 148-149°.¹⁷

2-(4-Nitrobenzoyl)-3-phenyl-1H-indole (**10a**) Yield: 40%, m.p 182-184°. IR (KBr, cm⁻¹): 3340 (N-H), 1613 (C=O), 770, 713 (Ar-H). ¹H-NMR (CDCl₃, , ppm): 7.1-8.1 (m, 13H, C₆H₅, COC₆H₄NO₂, *o*-disubstituted C₆H₄), 9.5 (hump, 1H, NH; D₂Oexchange). Anal. Calc. for C₂₁H₁₄N₂O₃: C, 73.68; H, 4.12; N, 8.18. Found: C, 73.10; H, 3.91; N, 8.11%.

2-Benzoyl-3-methyl-1H-indole(**11a**)Yield: 35%, m.p 138-140° as reported.¹⁶

2-(4-Bromobenzoyl)-3-methyl-1H-indole (**12a**) Yield: 25%, m.p 182-184°. IR (KBr, cm⁻¹): 3320 (N-H), 1611 (C=O), 835, 740 (Ar-H). ¹H-NMR (CDCl₃, , ppm): 2.3 (s, 3H, CH₃), 7-7.8 (m, 8H, COC₆H₄Br, and *o*-disubstituted C₆H₄), 9.0 (hump, 1H, NH; D₂Oexchange). Anal. Calc. for C₁₆H₁₂BrNO. 0.5H₂O: C, 59.46; H, 4.07; N, 4.34. Found: C, 59.49; H, 4.57; N, 4.57%. MS (FAB+), m/z: 314.08 (2.4%, M+H), 316.06 (2.2%, M+H+2).

2-(4-Chlorobenzoyl)-3-methyl-1H-indole (**13a**) Yield: 30%, m.p 167-168°. ¹H-NMR (CDCl₃, , ppm): 2.3 (s, 3H, CH₃), 7.3-8.1 (m, 8H, COC₆H₄Cl, and *o*-disubstituted C₆H₄), 9.3 (hump, 1H, NH; D₂Oexchange) as reported.¹⁷

Preparation of 1-methyl/ethyl-2-benzoyl/ tolyl-3- phenylindoles; 6c and 9b

The appropriate N-methyl/ethyl-o-aminophenylketone (**1b**, **c**), 0.02 mol) and the appropriate phenacyl bromide (**5a**,**e**; 0.02 mol) were treated similarly as under the general method for preparation of **6-13** (**a**).

2-Benzoyl-1-ethyl-3-phenyl-1H-indole (6c) Yield: 40%, m.p 82-84°. IR (KBr, cm⁻¹): 1633 (C=O), 747, 693 (Ar-H). ¹H-NMR (CDCl₃, , ppm): 1.5 (t, 3H, NCH₂CH₃), 4.5 (q, 2H, NCH₂CH₃), 7-7.9 (m, 14H, C₆H₅, COC₆H₅, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₃H₁₉NO: C, 84.89; H, 5.89; N, 4.30. Found: C, 85.03; H, 6.34; N, 4.34%.

1-Methyl-2-(4-methylbenzoyl)-3-phenyl-1H-indole (**9b**) Yield: 47%, m.p 103-104°. ¹H-NMR (CDCl₃, , ppm): 2.3 (s, 3H, COC₆H₄-CH₃), 4.1 (s, 3H, NCH₃), 7-8.1 (m, 13H, C₆H₅, COC₆H₄CH₃, *o*-disubstituted C₆H₄) Anal. Calc. for C₂₃H₁₉NO: C, 84.89; H, 5.89; N, 4.30. Found: C, 84.82; H, 6.32; N, 4.30%.

General procedure for preparation of Nalkyl/aralkyl- 3- phenyl-2-substitutedbenzoylindoles derivatives, (6-9 (b-f)

Powdered sodium hydroxide or potassium hydroxide (0.025 mol), indole derivative (6-13 (a); 0.01 mol) and tetrabutylammonium bromide (0.0025 mol) were mixed and triturated using magnetic stirrer for 10 min before adding the appropriate alkyl halide (0.011 mol). Ether (50 ml) was added to the reaction mixture after 6 hrs of trituration. Ethereal solution was then evaporated and the residue was either crystallized from methanol (6b-f, 7b-d, 8b-e and 9b-f) or purified using dry column flash chromatography (7e, f and 8f) where the elution system was hexane: ethyl acetate (9:1).

2-Benzoyl-1-methyl-3-phenyl-1H-indole

(**6b**) Yield: 82%, m.p 87-88° as reported.¹⁶

2-Benzoyl-1-ethyl-3-phenyl-1H-indole

(6c) Yield: 98%, m.p 82-84°. It was identical

with the sample prepared by the previouslymentioned method.

2-Benzoyl-3-phenyl-1-propyl-1H-indole (6d) Yield: 85%, m.p 83-85°. ¹H-NMR (CDCl₃, , ppm): 0.9 (t, 3H, NCH₂CH₂CH₃), 1.9 (m, 2H, NCH₂CH₂CH₃), 4.5 (t, 2H, NCH₂CH₂CH₃), 7.2-8.2 (m, 14H, C₆H₅, COC₆H₅, *o*-disubstituted C₆H₄) Anal. Calc. for C₂₄H₂₁NO: C, 84.92; H, 6.24; N, 4.13. Found: C, 84.84; H, 6.56; N, 4.30%.

2-Benzoyl-1-butyl-3-phenyl-1H-indole

(6e) Yield: 80%, m.p $58-60^{\circ}$. ¹H-NMR (CDCl₃, ppm): 0.9 (t, 3H, NCH₂CH₂CH₂CH₃), 1.2-2 (m, 4H, NCH₂CH₂CH₂CH₃), 4.6 (t, 2H, NCH₂CH₂CH₂CH₂CH₃), 7.2-8.1 (m, 14H, C₆H₅, COC₆H₅, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₅H₂₃NO: C, 84.95; H, 6.56; N, 3.96. Found: C, 84.86; H, 7.35; N, 4.11%.

2-Benzoyl-1-benzyl-3-phenyl-1H-indole (**6f**) Yield: 95%, m.p 119-120°. ¹H-NMR (CDCl₃, , ppm): 5.9 (s, 2H, CH₂C₆H₅), 7.2-8.1 (m, 19H, C₆H₅, COC₆H₅, *o*-disubstituted C₆H₄, CH₂C₆H₅). Anal. Calc. for C₂₈H₂₁NO: C, 86.71; H, 6.46; N, 3.61. Found: C, 86.14; H, 5.44; N, 3.63%.

2-(4-Bromobenzoyl)-1-methyl-3-phenyl-1H-indole (**7b**) Yield: 87%, m.p 133-135°. ¹H-NMR (CDCl₃, , ppm): 4.1 (s, 3H, NCH₃), 7.3-8.1 (m, 13H, C₆H₅, COC₆H₄Br, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₂H₁₆BrNO: C, 67.71; H, 4.13; N, 3.59. Found: C, 67.38; H, 4.34; N, 3.55%.

2-(4-Bromobenzoyl)-1-ethyl-3-phenyl-1Hindole (**7c**) Yield: 70%, m.p 108-110°. IR (KBr, cm⁻¹): 1631 (C=O), 739, 696 (Ar-H). ¹H-NMR (CDCl₃, , ppm): 1.5 (t, 3H, NCH₂CH₃), 4.6 (q, 2H, NCH₂CH₃), 7.3-8.1 (m, 13H, C₆H₅, COC₆H₄Br, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₃H₁₈BrNO: C, 68.26; H, 4.49; N, 3.46. Found: C, 67.69; H, 4.46; N, 3.27%.

2-(4-Bromobenzoyl)-3-phenyl-1-propyl-1H-indole (**7d**) Yield: 66%, m.p 141-143°. ¹H-NMR (CDCl₃, , ppm): 0.9 (t, 3H, NCH₂CH₂-CH₃), 1.8 (m, 2H, NCH₂CH₂CH₃), 4.5 (t, 2H, NCH₂CH₂CH₃), 7.2-8.1 (m, 13H, C₆H₅, CO-C₆H₄Br, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₄H₂₀BrNO.0.5H₂O: C, 67.39; H, 4.96; N, 3.34. Found: C, 67.64; H, 5.05; N, 3.24%.

2-(4-Bromobenzoyl)-1-butyl-3-phenyl-1Hindole (**7e**) Yield: 45%, oil. ¹H-NMR (CDCl₃, , ppm): 0.9 (t, 3H, NCH₂CH₂CH₂CH₃), 1.1-2.1 (m, 4H, NCH₂CH₂CH₂CH₃), 4.5 (t, 2H, NCH₂-CH₂CH₂CH₃), 7.2-8.1 (m, 13H, C₆H₅, COC₆H₄- Br, *o*-disubstituted C_6H_4). MS (FAB+), m/z: 431.31(2.2%, M), 433.19 (1.7%, M+2).

2-(4-Chlorobenzoyl)-1-methyl-3-phenyl-1H-indole (**8b**) Yield: 85%, m.p 118-20°. ¹H-NMR (CDCl₃, , ppm): 4.1 (s, 3H, NCH₃), 7.2-8.1 (m, 13H, C₆H₅, COC₆H₄Cl, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₂H₁₆ClNO: C, 76.41; H, 4.66; N, 4.05. Found: C, 76.07; H, 4.38; N, 4.04%.

2-(4-Chlorobenzoyl)-1-ethyl-3-phenyl-1Hindole (**8c**) Yield: 72%, m.p 58-60°. ¹H-NMR (CDCl₃, , ppm): 1.5 (t, 3H, NCH₂CH₃), 4.6 (q, 2H, NCH₂CH₃), 7.2-8.1 (m, 13H, C₆H₅, CO-C₆H₄Cl, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₃H₁₈ClNO: C, 76.77; H, 5.64; N, 3.89. Found: C, 76.45; H, 6.38; N, 3.71%.

2-(4-Chlorobenzoyl)-3-phenyl-1-propyl-1H-indole (**8d**) Yield: 68%, m.p 135-137°. ¹H-NMR (CDCl₃, , ppm): 0.9 (t, 3H, NCH₂-CH₂CH₃), 1.9 (m, 2H, NCH₂CH₂CH₃), 4.5 (t, 2H, NCH₂CH₂CH₃), 7.2-8.2 (m, 14H, C₆H₅, COC₆H₅, *o*-disubstituted C₆H₄) Anal. Calc. for C₂₄H₂₀CINO: C, 77.10; H, 5.39; N, 3.75. Found: C, 76.59; H, 5.68; N, 3.73%.

1-Butyl-2-(4-chlorobenzoyl)-3-phenyl-1Hindole (**8e**) Yield: 44%, m.p 55-56°. ¹H-NMR (CDCl₃, , ppm): 0.9 (t, 3H, NCH₂CH₂-CH₂C<u>H₃</u>), 1.9 (m, 4H, NCH₂C<u>H₂CH₂CH₃</u>), 4.6 (t, 2H, NCH₂CH₂CH₂CH₂), 7.2-8.1 (m, 13H, C₆H₅, COC₆H₄Cl, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₅H₂₂ClNO: C, 77.41; H, 5.72; N, 3.61. Found: C, 77.02; H, 5.70; N, 3.77%.

1-Methyl-2-(4-methylbenzoyl)-3-phenyl-1H-indole (**9b**) Yield: 95%, m.p 103-104°. It is to be noted down that m.p as well as spectral and elemental analyses were all identical with the sample prepared by the previouslymentioned method.

1-Ethyl-2-(4-methylbenzoyl)-3-phenyl-1Hindole (**9c**) Yield: 90%, m.p 75-76°. IR (KBr, cm^{-1}): 1632 (C=O), 741, 697 (Ar-H). ¹H-NMR (CDCl₃, , ppm): 1.5 (t, 3H, NCH₂CH₃), 2.3 (s, 3H, $COC_6H_4CH_3$), 4.6 (q, 2H, NCH_2CH_3), 7-8.1 (m, 13H, C_6H_5 , $COC_6H_4CH_3$, *o*-disubstituted C_6H_4). Anal. Calc. for $C_{24}H_{21}NO$: C, 84.92; H, 6.24; N, 4.13. Found: C, 85.22; H, 6.34; N, 4.13%.

2-(4-Methylbenzoyl)-3-phenyl-1-propyl-1H-indole (**9d**) Yield: 90%, m.p 110-112°. ¹H-NMR (CDCl₃, , ppm): 0.9 (t, 3H, NCH₂-CH₂CH₃), 1.9 (m, 2H, NCH₂CH₂CH₃), 2.3 (s, 3H, COC₆H₄CH₃), 4.5 (t, 2H, NCH₂CH₂CH₃), 7-8.1 (m, 13H, C₆H₅, COC₆H₄CH₃, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₅H₂₃NO: C, 84.95; H, 6.56; N, 3.96. Found: C, 84.91; H, 6.89; N, 4.29%.

1-Butyl-2-(4-methylbenzoyl)-3-phenyl-1H-indole (**9e**) Yield: 65%, m.p 110-112°. ¹H-NMR (CDCl₃, , ppm): 0.9 (t, 3H, NCH₂-CH₂CH₂CH₃), 1-2.2 (m, 4H, NCH₂CH₂-CH₂CH₃), 2.3 (s, 3H, COC₆H₄CH₃), 4.5 (t, 2H, NCH₂CH₂CH₃), 7-8.1 (m, 13H, C₆H₅, CO-C₆H₄CH₃, *o*-disubstituted C₆H₄). Anal. Calc. for C₂₆H₂₅NO: C, 84.98; H, 6.86; N, 3.81. Found: C, 84.86; H, 7.41; N, 4.14%.

1-Benzyl-2-(4-methylbenzoyl)-3-phenyl-1H-indole (**9f**) Yield: 55%, m.p 92-93°. ¹H-NMR (CDCl₃, , ppm): 2.3 (s, 3H, CO- $C_6H_4CH_3$), 5.8 (s, 2H, CH₂C₆H₅), 7-8 (m, 18H, C_6H_5 , COC₆H₄CH₃, *o*-disubstituted C₆H₄, CH₂C₆H₅). Anal. Calc. for C₂₉H₂₃NO: C, 86.67; H, 5.77; N, 3.49. Found: C, 86.14; H, 6.36; N, 3.38%.

Synthesis of 1,2-Dibenzoyl-3-phenyl-1Hindole (6g)

To a suspension of sodium hydride (60%, 0.012 mol, 0.48 g in anhydrous DMF (20 ml) was added dropwise a solution of 2-benzoyl-3phenyl-1H-indole (6a, 0.01 mol, 2.97 g) in DMF (20 ml), keeping the temperature below 25°. When the addition was complete, the reaction was heated to 40° until hydrogen evolution ceased. After cooling to 25°, a solution of benzoyl chloride (0.01 mol, 1.16 ml) in DMF (10 ml) was added dropwise, keeping the temperature below 25°. The mixture was stirred for 2 hrs and then poured onto ice-water (100 ml). The resulting solid was filtered and crystallized from methanol. Yield: 88%, m.p 206-208°. IR (KBr, cm⁻¹): 1697, 1629 (2 bands, C=O), 748, 723, 690 (Ar-H). ¹H-NMR (CDCl₃, , ppm): 7.2-8.1 (m, 19H, C, H) = 0 displaying the constant of th C_6H_5 , 1, 2-di-COC₆H₅, \hat{o} -disubstituted C_6H_4). Anal. Calc. for C₂₈H₁₉NO₂: C, 83.69; H, 4.77; N, 3.49. Found: C, 83.07; H, 5.34; N, 3.49%.

Pharmacology

The newly synthesized indole derivatives (6a-g, 7a-f, 8a-f, 9a-13a) were tested for their analgesic effect using hot-plate test in mice. Compounds 7b and 8b were tested for their antipyretic and anti-inflammatory effects using veast induced hyperthermia and paw edema in rats. Both compounds were also tested for their ulcerogenic activity in mice. All compounds were administered to the animals as a solution in 4% tween 80. Pure standard indomethacin (Nile co.) was used as a reference in the same dose level. Male albino mice (30-35 g) and male rats (120-150 g) were used in the biological studies. They were housed in groups of 5 per cage, fed ad libitum with rodent's chow and allowed free access to drinking water. Data were analyzed using the student's t-test and the values were expressed as mean \pm S.E. (n=5 animals).

Analgesic Activity

Male albino mice (30-35 g) were placed for testing on a hot-plate apparatus, the surface temperature of which was maintained at 55 \pm 0.5°. The reaction time in seconds was taken as the time period from the instant the animal reached the hot plate until the moment the animal licked its hind paw or jumped out within a plexiglass cylinder placed on the hotplate.¹⁸ Animals were randomly divided into groups of 5 mice each and each animal was used once.. Testing was done after 1 hr of the intrapretoneal injection of tested compounds (5, 10 mg/kg), indomethacin (5, 10 mg/kg) or vehicle (control). Mean licking time \pm S.E. was evaluated for each group and results were represented in Fig.2.

Antipyretic activity

Male albino rats (120-150 g) were made febrile by subcutaneous injection into the dorsal region of brewer's yeast suspension (1 ml/100 g, 20%) in normal saline at 37° following Adams et al method.¹⁹ Then, animals were fasted, but allowed access to water until the termination of experiments that were performed in a temperature-controlled room. Six hours later, those rats which developed a satisfactory pyrexia (at least 1° higher than normal body temperature 36.8 \pm 0.21) were randomly divided into groups of 5 animals each and received i.p., respectively, tested compounds (7b, **8b**) (5, 10 mg/kg), vehicle indomethacin (5, 10 mg/kg) or (control). The rectal temperature was recorded on a thermometer inserted 4 cm in the rectum for 45 sec before and 1, 2, 3, 4, and 5 hrs after dosing. Results are represented in Fig. 3, 4.



Fig. 2: Mean licking time of 6a-13a, 6b-g, 7b-f & 8b-f in the hot-plate test.
Significantly different from control (vehicle) at *p<0.05, **p<0.01



Fig. 3: Antipyretic activity of **7b** & **8b** (5 mg/kg) using the yeast induced hyperthermia test. Significantly different from control at *p< 0.05, **p< 0.01.



Fig. 4: Antipyretic activity of **7b** & **8b** (10 mg/kg) using the yeast induced hyperthermia test.

Significantly different from control at *p< 0.05, **p< 0.01.

Anti-inflammatory Activity

The anti-inflammatory activity was done according to the method described by Winter *et al.*,²⁰ where a pedal inflammation in rat paws induced by subplantar injection of brewer's yeast suspension (1 ml/100 g, 20%) into the right hind of the rats.

Male albino rats (120-150 g) were randomly divided into groups of 5 animals each. The thickness of rat paw was measured by a Vernier caliper (SMIEC) before and 20 hr after yeast injection to detect the inflammation induced by yeast. Tested compounds (**7b** and **8b**) at doses of 5 and 10 mg/kg were injected i.p. to four different groups of rats. Control group received a vehicle (tween 80 solution), while reference groups received indomethacin i.p. at a dose 5 and 10 mg/kg.

The difference between the thickness of the two paws was taken as a measure of edema. The measurement was carried out at 0.0, 0.5, 1, 2, 3, 4, 5, 12 hrs after injection of the tested compounds, reference and vehicle. The percentage of edema and percentage of inhibition were calculated²¹ and results are listed in Tables 1 and 2.

Gastric ulceration

Male mice were fasted for 24 hrs. Tested compounds were administered orally to groups of 6 mice. After 6 hrs, the animals were scarified, the stomachs were removed and gastric lesions on the mucosa were determined by using stereoscopic microscope.²² "Ulcer" was defined as at least one lesion that was 0.5 mm or more in length. All lesions of more than 0.1 mm in length were summed to obtain the ulcer index. Results are listed in Table 3.

Group	Dose mg/kg	Percentage of edema (%)							
1		T ₀	0.5 hr	1hr	2 hr	3 hr	4 hr	5 hr	12 hr
Control		60.97	63.31	63.57	65.47	66.01	66.16	66.96	66.96
Indomethacin	5	55.52	53.47	52.00	52.00	46.94	47.31	48.91	59.02
	10	55.20	52.40	52.40	50.87	43.59	43.59	44.61	40.50
7b	5	57.86	52.91	49.96	47.91	41.27	42.38	42.81	58.66
	10	55.63	49.80	47.56	41.08	36.41	38.21	38.96	50.29
8b	5	57.15	52.46	50.93	49.30	43.66	44.08	45.92	59.44
	10	57.26	52.92	50.64	48.48	45.55	41.60	41.82	16.84

Table 1: Effects of compounds 7b and 8b on yeast induced paw edema in rats.

Group	Dose mg/kg	Percentage of edema inhibition								
		T_0	0.5hr	1hr	2hr	3hr	4hr	5hr	12hr	
Indomethacin	5	20.75	33.96	38.46	43.35	62.05	54.43	53.15	29.56	
	10	21.17	36.24	36.96	45.42	60.20	60.46	60.25	66.39	
7b	5	12.36	35.10	42.96	51.64	63.91	62.47	63.16	30.21	
	10	12.36	35.10	42.96	51.64	63.91	62.47	63.16	30.21	
8b	5	14.88	36.24	40.71	48.87	60.20	59.79	58.31	27.94	
	10	14.04	34.72	41.08	50.25	56.82	63.48	64.45	89.98	

Table 2: Inhibitory effect of compounds 7b and 8b on yeast induced paw edema in rats.

Table 3: Effect of compounds **7b** and **8b** on
gastric mucosa of mice.

Group	Dose mg/kg	Ratio of ulcerated animals	Ulcer index (mean ± S.E.)			
	5	2/6	0.85 ± 0.07			
Indome- thacin	10	4/6	1.76±0.12			
	30	6/6	2.10±0.17			
	50	Not tested				
7b	5	0/6	0.00			
	10	0/6	0.00			
	30	2/6	1.50 ± 0.07			
	50	4/6	1.50 ± 0.07			
8b	5	0/6	0.00			
	10	2/6	1.35 ± 0.07			
	30	3/6	1.38±0.02			
	50	5/6	1.66±0.22			

RESULTS AND DISCUSSION

Chemistry

Monoalkylation of o-aminoketones;(1b-c, 2b-c)

Polyalkylation is the major drawback limiting the use of direct alkylation of primary amines in synthesis of secondary amines. Different methods were developed to solve this problem.^{15,23} In the present work, it seemed reasonable to affect monoalkylation of *o*-

aminoketones through tosylation of these amines¹⁵ followed by *N*-alkylation¹⁵ and hydrolysis of *N*-alkyl-*N*-tosylated derivatives,²⁴ Fig. 1.

Preparation of N-alkyl-3-phenyl-2aroylindoles; (6c, 9b)

Cyclisation of the prepared *N*-alkyl-*o*aminobenzophenones (**1b**, **c**) with the appropriate phenacyl bromide (**5a,b**) to give 1alkyl-3-phenyl-2-aroylindoles was attempted, Fig. 1. This approach proved to be of limited utility due to formation of non-purifiable resinous products, low overall yield and being a multistepped method. Therefore, another approach was successfully adopted.

Synthesis of 3-methyl/phenyl -2-aroylindoles (6a-13a)

Compounds (**6a**, **8a**, **9a**, **11a** and **13a**) are reported to be prepared by treating the appropriately substituted o-amino-phenyl ketone with -haloketones.^{16,17} In the present study the same method was adopted to prepare similar compounds (**7a**, **10a** & **12a**) as confirmed by elemental and spectral analyses. EI spectra of compounds (**7a**, **12a**) didn't show the molecular ion peak. However, fast atomic bombardment (FAB) technique revealed the presence of quasimolecular ion peaks: **7a**; ($M^{+*}+H+2$) at m/z 378.9 corresponding to ($C_{21}H_{14}^{81}BrNO+H$) and ($M^{+*}+H$) at m/z 376.9 corresponding to $(C_{21}H_{14}^{79}BrNO+H))$ and 12a; $(M^{+\bullet}+H+2)$ at m/z 316 corresponding to $(C_{16}H_{12}^{81}BrNO+H)$ and $(M^{+\bullet}+H)$ at m/z 314 corresponding to $(C_{16}H_{12}^{79}BrNO+H)$.

Synthesis of 1-alkyl-2-aroyl-3-phenylindoles (6b-f, 7b-f, 8b-f, 9b-f)

N-Alkylation of aroylindoles was successfully achieved adopting solid /liquid phase transfer catalysis SL-PTC^{25,26} using tetrabutylammonium bromide (TBAB) as a without solvent. Structural catalyst and elucidation of the prepared compounds was confirmed by elemental and spectral analyses. Although, the molecular ion peak was absent in EI spectrum of compound 7e, it could be recognized through FAB spectrum by the presence of ion peaks; (M⁺+2) at m/z 433 corresponding to $(C_{25}H_{22}^{-81}BrNO)$ and (M^{+}) at m/z 431 corresponding to $(C_{25}H_{22}^{-79}BrNO)$. It is to be noted down that PTC seems to be a superior technique in terms of yield, purity of products and non-tediousness of reaction.

Synthesis of 1,2-Dibenzoyl-3-phenylindole (6g)

N-Acetylation of indoles is reported to proceed using acetic anhydride/ K_2CO_3 .^{27, 28} In the present work, *N*-benzoylation of 6a to yield 6g was achieved using sodium hydride and benzoyl chloride.

Pharmacology

Analgesic activity

The effect of tested compounds or reference drug on mean licking time in hotplate test was used as a measure of analgesic activity. It was compared at 5 and 10 mg/kg dose levels with a control group which was injected with vehicle. Results presented in Fig.2 revealed that; among the 2-aroyl-3phenylindoles (6a-10a),compounds that showed significant analgesic activity were those (un)substituted by halogen at the para position of 2-aroyl entity (-, Br, Cl; 6a, 7a, 8a respectively). Compounds substituted bv methyl or nitro were found to be inactive in regard to both doses. Also, 2-aroyl-3methylindoles (11a-13a) were devoid of analgesic activity at both doses.

The effect of substitution at nitrogen atom of certain indoles on analgesic activity was examined. Results indicate that N-substitution on the indole ring by groups such as methyl, propyl, butyl, benzyl or benzoyl in compounds **6**, **7**, **8** (**b**, **d**, **e**, **f**) and **6g**, increases the analgesic activity, with the exception of N-ethyl derivatives; (**6c**, **7c** and **8c**). The highest activity was found to reside in *N*-methyl derivatives **7** and **8** (**b**); Fig. 2.

Antipyretic activity

Investigation of the effects of tested compounds (**7b**, **8b**) compared with indomethacin on yeast induced hyperthermia test, represented in Fig. 3, 4, indicates that they exhibit significantly marked reduction in rectal temperature. Also, animals reached normal temperature after 1 hr in case of compounds **7b** and **8b** and within 3-4 hrs in case of indomethacin

Anti-inflammatory activity

Effects of tested compounds (7b, 8b) and indomethacin on yeast induced inflammation were represented as the time course of edema, (Table 1). At 5 mg/kg-dose level, results tested indicated that compounds and indomethacin caused a marked decrease in edema if compared to control group with slight faster and more effective reduction in case of tested compounds. Also, a decrease in activity was noticeable after 5 hrs of administration of both tested compounds and indomethacin. At 10 mg/kg-dose level, a decrease in the activity of compound (7b) was noticed after 5 hrs while compound (8b) persisted in reducing edema during 12 hrs period and making an optimum reduction at the end of the period (edema% =10%). In a similar way, the comparison may utilize the inhibitory effects of tested compounds and indomethacin on the yeast induced edema versus time as represented in Table 2.

Gastric ulceration

Occurrence of gastrointestinal damage (bleeding, ulceration) is probably the most prevalent and serious side effect associated with the use of NSAIDs.²⁹ Observation of the gastrointestinal mucosa for the presence of lesions following oral administration of a single dose of the tested compounds or reference drug has been taken as an indication for the ulcerogenic effects. Both the frequency of ulceration (expressed as ratio of ulcerated

animals) and the severity of ulceration (expressed as ulcer index) were used for comparison of the tested compounds and indomethacin. Results represented in Table 3 indicated that compound **7b** has no ulcerogenic toxicity at treating doses (5, 10 mg/kg) while compound **8b** caused mild ulceration in one third of the animals at 10 mg/kg. At higher doses both tested compounds exhibited lower ulcerogenic activity compared with indomethacin which caused ulceration at all doses increased in frequency and severity as increasing doses.

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