

SYNTHESIS AND *IN-VITRO* CYTOTOXIC ACTIVITY OF NOVEL BENZO[b]PHENAZINE-6,11-DIONE AND 1,4-NAPHTHOQUINONE DERIVATIVES

Maha M. A. Khalifa^{1*}, Magda M. F. Ismail¹ and Eman Noaman²

¹*Department of Pharmaceutical Chemistry, Faculty of Pharmacy (Girls), Al-Azhar University, Nasr City, Cairo, Egypt*

²*Department of Radiation Biology, Natural Center for Radiation Research and Technology, Atomic Energy Authority, Cairo, Egypt*

تم تحضير مشتقات من البنزو [ب] فينا زين - دايون و - نافثوكينون الجديدة وذلك بواسطة - داي كلورو - نافثوكينون و داي اريل أمين/ فينلين الدامين. تم اثبات البناء الكيماوى لها عن طريق التحليل الدقيق للعناصر والأشعة تحت الحمراء ، الرنين النووى المغناطيسى وإشعة الكتلة. وقد تم دراسة التسمم الخلوي للمركبات وذلك بزرع الخلايا السرطانية في الفئران وقد وجد تأثير للمركبين ضد السرطان وكذلك وجد أن ا فاعلية في خط خلايا بشرية ضد سرطان الرئة.

5,12-Dihydrobenzophenazine-6,11-diones, 2-Arylamino-3-chloro-1,4-naphthoquinones and 6,11-dihydrobenzo[b]phenazine-6,11-diones, were synthesized from 2,3-dichloro-1,4-naphthoquinone and arylamines/phenylenediamines. Studying the cytotoxicity using EAC and human cell lines revealed that 5,12-dihydrobenzo[b]phenazine-6,11-dione (3) and 3-chloro-2-(2-pyridylamino)-1,4-naphthoquinone (10) showed selective cytotoxicity against the human lung carcinoma cell line (H460) superior to doxorubicin. Compound 3 (16.25 μ M) was 1.3 times higher than that of doxorubicin. However, IC50 value of compound 10 was 9.90 μ M which was 2 times higher than that (20.10 μ M) of doxorubicin. These compounds were inactive against liver carcinoma (HEPG2), brain tumor (U251), cervix carcinoma (HELA) and breast carcinoma (MCF7) cell lines.

Received in 17/3/2008, Received in revised form in 17/6/2008 & Accepted in 19/6/2008

*Corresponding author e-mail address: maha_khalifa@hotmail.com.

INTRODUCTION

Studies on the activity of heterocyclic quinones containing nitrogen showed that the number and position of nitrogens are considerably important for cytotoxicity¹. The diazanaphthoquinone were proved to be the most active compounds in comparison with naphthoquinone and quinolindione². Another structural requirements for the antitumor activity is the p-quinone moiety in the non heterocyclic ring, however o-quinone gave decreased activity^{3&4}. One of the proposed mechanisms of coplanar polycyclic compounds is that they act as topoisomerase inhibitor via DNA intercalation⁵⁻⁷. The topoisomerase are essential enzymes in the regulation of DNA topology which is required if cells are to divide and proliferate⁸. The phenazines have been shown to fulfill the fundamental physicochemical requirements for DNA intercalation⁹. However it was reported by Johnson and approved by Lee¹⁰ that the antitumor activity is enhanced as more heterocyclic rings were annulated to the heteroquinone ring as well as its dependency on the number of nitrogen atoms. Most of the previously reported phenazines contain unsaturated nitrogen atom^{9&10}, however in the present investigation, a novel series of substituted 5,12-dihydrobenzo-

[b]phenazine-6,11-diones and their bioisosteres^{11&12} having planar ring and *p*-conjugated ketone groups containing a nitrogen atom which may enable additional hydrogen bonding with DNA was synthesized.

Prompted by the fact that streptonigrin which is an alkaloid having excellent antitumor activities¹³. We also synthesized 2-Arylamino-3-chloro-1,4-naphthoquinones and their corresponding rigid structures. All the compounds were then submitted for preliminary *in vitro* screening for cytotoxic activity.

EXPERIMENTAL

Chemistry

Melting points were determined on electrothermal 9100 digital melting point apparatus and were uncorrected. ¹HNMR spectra were recorded in DMSO-d₆ on Varian Gemini 200 (200 MHz) using tetramethylsilane (TMS) as an internal standard (chemical shift in δ ppm). The IR spectra were performed on a Perkin-Elmer 1600 FTIR in KBr pellets. Elemental microanalysis (C,H,N) were performed on a Perkin-Elmer 2400 analyzer from vacuum-dried samples at the micro analytical unit of Cairo University. All compounds were within $\pm 0.4\%$ of the theoretical values. The mass spectra were recorded on a Hewlett-Packard 5988-A instrument at 70

eV. Chemicals were purchased from E. Merck (Darmstadt, Germany), Sigma-Aldrich (Germany); solvents used were of the highest grade.

General procedure for the preparation of 5,12-dihydro-substitutedbenzo[*b*]phenazine-6,11-diones (3-6)

o-Phenylene diamine derivatives, **2** (0.5 mmol) was added to a solution of 2,3-dichloro-1,4-naphthoquinone **1** (0.5 mmol) in DMF (20 mL) and heated under reflux for 5 h. The reaction mixture was allowed to cool, poured into ice-water and the product obtained was filtered, dried and recrystallized from appropriate solvent (Table 1).

5,12-Dihydrobenzo[*b*]phenazine-6,11-dione (3). IR (KBr, cm^{-1}): 3423 (NH), 1676 (CO), $^1\text{HNMR}$ (DMSO- d_6): δ 6.90-7.33 (m, 4H, H_1 , H_2 , H_3 , H_4), 7.47-8.16 (m, 4H, naphthoquinone -Hs), 9.20 (s, 2H, 2 NH). MS (m/z %): 280 (M+ H_2O , 100), 252 (24.9), 217 (29.2), 188 (11.4).

2-Methyl-5,12-dihydrobenzo[*b*]phenazine-6,11-dione (4). IR (KBr, cm^{-1}): 3270 (NH), 1666 (CO), $^1\text{HNMR}$ (DMSO- d_6): δ 2.64 (s, 3H, CH_3), 7.27 (s, 1H, H_1), 7.79-7.93 (m, 2H, H_3 , H_4), 8.06-8.21 (m, 4H, naphthoquinone -Hs), 9.20 (s, 1H, NH), 9.28 (s, 1H, NH). MS (m/z %): 294 (M+ H_2O , 100), 229 (29.8), 147 (17.5), 102 (17.5).

2,3-Dimethyl-5,12-dihydrobenzo[*b*]phenazine-6,11-dione (5). IR (KBr, cm^{-1}): 3244 (NH), 1654 (CO), $^1\text{HNMR}$ (DMSO- d_6): δ 2.08 (s, 6H, 2 CH_3), 6.26 (s, 2H, H_1 , H_4), 7.28-8.40 (m, 4H, naphthoquinone-Hs), 10.66 (s, 2H, 2 NH).

2-Chloro-5, 12-dihydrobenzo[*b*]phenazine-6,11-dione (6). IR (KBr, cm^{-1}): 3320 (NH), 1672 (CO), $^1\text{HNMR}$ (DMSO- d_6): δ 7.89-7.96 (m, 3H, H_1 , H_3 , H_4), 8.05-8.12 (m, 4H, naphthoquinone-Hs), 9.02 (s, 2H, 2 NH).

2-Chloro-12*H*-benzo[*b*]phenoxazine-6,11-dione (7)

To a solution of 2,3-dichloro-1,4-naphthoquinone **1** (0.5 mmol) in EtOH (30 mL), add KOH (0.39 g, 0.01 mol) and 4-chloro-*o*-aminophenol (0.5 mmol) Then the mixture was heated under reflux for 6h, allowed to cool, poured into ice-water and the product obtained was filtered, dried and recrystallized from appropriate solvent (Table 1). **7**: IR (KBr, cm^{-1}): 3180 (NH), 1640 (CO), $^1\text{HNMR}$ (DMSO- d_6): δ 6.53-6.65 (m, 2H, H_3 , H_4), 7.65-7.88 (m, 2H, H_8 , H_9), 8.01-8.06 (m, 3H, H_1 , H_7 , H_{10}), 9.14 (s, 1H, NH, D_2O exchangeable). MS (m/z %): 299 (M+2, 30.1), 297 (M $^+$, 50.4), 281 (85.8), 253 (51.3), 220 (56.6), 50 (100).

Table 1: Physical properties and molecular formula of the synthesized compounds.

Compd No.	M.P. (°C)	Solvent of Cryst.	Yield (%)	Mol.Formula M.Wt	Calcd.	Found
3	198-200	EtOH	90	$C_{16}H_{10}N_2O_2 \cdot H_2O$ 280.26	C 68.57 H 4.28 N 9.99	68.99 3.85 9.99
4	266-68	EtOH	90	$C_{17}H_{12}N_2O_2 \cdot H_2O$ 294.29	C 69.31 H 4.75 N 9.51	69.72 4.56 9.91
5	246-48	Pet ether/ CH_2Cl_2	80	$C_{18}H_{14}N_2O_2 \cdot 0.5H_2O$ 299.32	C 72.16 H 5.01 N 9.35	72.10 4.90 9.35
6	255-57	EtOH	50	$C_{16}H_9ClN_2O_2$ 296.71	C 64.77 H 3.06 N 9.43	64.50 3.20 9.26
7	270-72	EtOH	40	$C_{16}H_8ClNO_3$ 297.69	C 64.55 H 2.71 N 4.71	64.80 2.90 4.95
8	270-72	EtOH/Ether	40	$C_{16}H_9NO_2S$ 279.31	C 68.80 H 3.25 N 5.01	68.99 3.20 5.57
10	143-46	EtOH	80	$C_{15}H_9ClN_2O_2 \cdot 0.5 H_2O$ 293.70	C 61.22 H 3.74 N 9.53	61.25 3.93 9.45
11	193-95	EtOH/Ether	65	$C_{18}H_{14}ClNO_2 \cdot 0.25 H_2O$ 316.26	C 68.29 H 4.58 N 4.42	68.63 4.43 4.37
12	286-88	EtOH	30	$C_{15}H_7N_3O_2$ 261.20	C 68.96 H 2.70 N 16.07	68.68 3.01 16.55
13	186-88	EtOH/Ether	42	$C_{18}H_{12}N_2O_2$ 288.30	C 74.99 H 4.20 N 9.71	75.17 4.14 9.79

**2-Chloro-12H-benzo[*b*]pheno-
thiazine-6,11-dione (8)**

The above procedure was followed using 4-chloro-*o*-aminothiophenol (0.5 mmol) and the product obtained was purified using column chromatography using cyclohexane : EtOAc (1:3) as an eluant. IR (KBr, cm^{-1}): 3190 (NH), 1640 (CO), $^1\text{H NMR}$ (DMSO- d_6): δ 7.4 (t, 1H, H₁), 7.6 (t, 1H, H₃), 7.69-7.80 (m, 2H, H₁, H₄), 7.85-7.90 (t, 2H, H₈, H₉), 8.10 (d, 2H, H₇, H₁₀), 9.20 (s, 1H, NH).

General procedure for the preparation of 2-arylamino-3-chloro-1,4-naphthoquinones (10 and 11)

2-aminopyridine/2-ethylaniline (0.5 mmol) was added to a solution of 2,3-dichloro-1,4-naphthoquinone **1** (1.135 g, 0.5 mmol) in ethanol (30 mL) and heated under reflux for 3 h. The reaction mixture was cooled and then filtered. The product was crystallized from the appropriate solvent (Table 1).

3-chloro-2-(2-pyridylamino)-1,4-naphthoquinone (10). IR (KBr, cm^{-1}): 3327 (NH), 1678 (CO), 1584 (C=N). $^1\text{H NMR}$ (DMSO- d_6): δ 7.89-7.96 (m, 4H, H₃, H₄, H₅, H₆), 8.06-8.09 (m, 4H, naphthoquinone-Hs), 9.20 (s, 1H, NH, D₂O exchangeable).

3-Chloro-2-(2-ethylphenylamino)-1,4-naphthoquinone (11). IR (KBr, cm^{-1}): 3328 (NH), 1678 (CO);

$^1\text{H NMR}$ (DMSO- d_6): δ 1.13 (t, 3H, CH₃, $J= 7.5$ Hz), 2.63 (q, 2H, CH₂, $J= 7.5$ Hz), 7.14-7.25 (m, 4H, H₃, H₄, H₅, H₆), 7.78-8.08 (m, 2H, H₆, H₇), 8.09-8.11 (m, 2H, H₅, H₈), 9.01 (s, 1H, NH).

General procedure for the preparation of 1, 5, 12-Triaza-naphthacene-6,11-dione and 6,11-Dihydro-1-ethylbenzo[*b*]phenazine-6,11-dione (12 and 13)

A mixture of 0.5 mmol of **10** or **11** in 50 mL of DMF and (0.65 g, 0.01 mol) of sodium azide, suspended in a little amount of water (1 mL) was heated on the steam bath overnight. The reaction was chilled, the filtered precipitate was extracted with methylene chloride and concentrated, and then the residue was purified by crystallization (Table 1).

1,5,12-Triaza-naphthacene-6,11-dione (12). IR (KBr, cm^{-1}): 1626 (C=N), 1687 (CO). $^1\text{H NMR}$ (DMSO- d_6): δ 7.39 (t, 1H, H₃), 7.59 (d, 1H, H₄), 7.81 (t, 2H, H₈, H₉), 7.97-7.99 (m, 2H, H₇, H₁₀), 8.13 (d, 1H, H₂).

6,11-Dihydro-1-ethylbenzo[*b*]phenazine-6,11-dione (13). IR (KBr, cm^{-1}): 1585 (C=N), 1677 (CO). $^1\text{H NMR}$ (DMSO- d_6): δ 1.38 (t, 3H, CH₃, $J= 7.5$ Hz), 3.35 (q, 2H, CH₂, $J= 7.5$ Hz), 7.95-8.06 (m, 4H, H₂, H₃, H₈, H₉), 8.20 (d, 1H, H₄), 8.30-8.33 (m, 2H, H₇, H₁₀). MS (m/z %): 288 (M^+ , 17.0), 248 (58.2), 232 (23.5), 221 (47.1).

Methods of antitumor screening Activity against EAC experimental cell line

Animals, chemicals and facilities: Female Swiss albino mice weighing 25-30 g obtained from (the holding company of biological products and vaccines, VACSERA, Cairo, Egypt) were housed at a constant temperature ($24\pm 2^{\circ}\text{C}$) with alternating 12 h light and dark cycles and fed standard laboratory food (Milad Co., Cairo Egypt) and water *ad libitum*. All chemicals and reagents were from Sigma-Aldrich Germany and Merck- Germany.

Ehrlich Ascites Carcinoma cells (EAC) were obtained by needle aspiration of ascetic fluid from preinoculated mice; under aseptic conditions. Tumor cells suspension (2.5×10^6 per mL) was prepared. Tested compounds were prepared with various dilutions in DMSO (1 mL). In a set of sterile test tubes 0.8 mL MBIR-1640, 0.1 mL of each of the tested compounds (corresponding to 0.34, 0.265, 0.177, 0.088 and $0.035\ \mu\text{M}/\text{mL}$) and 0.8 mL of media (RBMI-contain glutamine and fetal calf serum as nutrient beside penicillin and streptomycin as antibiotics) were mixed then, 0.1 mL of tumor cell suspension was added. The test tubes were incubated at 37°C for 2 h. Then, trypan blue exclusion test¹⁴ was carried out to calculate the percentage of non-viable cells.

Activity against human cell lines

Cells were plated in 96-multiwell plate (10^4 cells/well) for 24 h before treatment with the compounds to allow attachment of cell to the wall of the plate. Different concentrations of the compounds under test ($0\text{-}36.00\ \mu\text{M}/\text{mL}$) were solubilised in dimethyl-sulfoxide (DMSO) and were added to the cell monolayer of the five human cell lines. Monolayer cells were incubated with the compounds for 48 h at 37°C and in atmosphere of 5% CO_2 . After 48 h, cells were fixed, washed and stained with sulforhodamine B stain. The color intensity was measured in an ELISA reader¹⁵. The relation between surviving fraction and drug concentration is plotted to get the survival curve for the active compounds. Statistical Analysis Student's t test was used for analysis of the biochemical parameters. The data were expressed as mean \pm standard error¹⁶.

RESULTS AND DISCUSSION

Chemistry

The preparation of target compounds was conducted according to the sequence of reactions are depicted in Schemes 1 and 2. Treatment of **1** with 1,2-phenylenediamines, **2** in DMF under reflux condition² afforded the desired compounds, **3-6**. In analogy,

reaction of 2-amino-4-chlorophenol or thiophenol with **1** using KOH in refluxing ethanol for furnished the desired bioisosteres, **7** and **8**.

2-Arylamino-3-chloro-1,4-naphthoquinones, **10**, **11** were obtained by reacting 2,3-dichloro-1,4-naphthoquinone, **1** with arylamines **9** in ethanol. The latter were reacted with sodium azide in DMF at 90-100°C overnight¹⁰ to give our target compounds, 1,5,12-Triaza-naphthacene-6,11-dione and 6,11-dihydro-1-ethylbenzo[*b*]phenazine-6,11-dione respectively (**12**, **13**). The reaction is believed to proceed via the formation of the unstable intermediate, the 2-azido derivative¹⁰.

Cytotoxic activity

Activities against Ehrlich Ascites Carcinoma (EAC) cell line

These biological studies were performed at the National Center for Radiation Research and Technology (NCRRT), Cairo, Egypt. A preliminary screening on the new compounds was performed against EAC cells¹⁷. The tumor cell suspensions were incubated with different concentrations in mM/mL^{15&18} (Table 2).

In vitro antitumor screening of 5,12-dihydrobenzo[*b*]phenazine-6,11-diones (**3-6**) and their classic bioisosteres (**7**, **8**) revealed that, the unsubstituted derivative, **3** exhibited higher cytotoxic activity (IC₅₀ = 0.035 mM) compared to that of the standard. The presence

of electron-donating substituent (CH₃) at position 2, decreased the activity (**4**, IC₅₀ = >0.354 mM). Introduction of another CH₃ group at position 3 in compound **5**; (IC₅₀ = >0.354 mM) showed also decrease in cytotoxic activity. On the other hand, introduction of electron-withdrawing group (Cl) at position 2 of benzo[*b*]phenazine nucleus compound **6** enhanced its cytotoxicity (IC₅₀ = >0.035 mM) which may enable nitrogen at position 5 of hydrogen bonding with DNA.

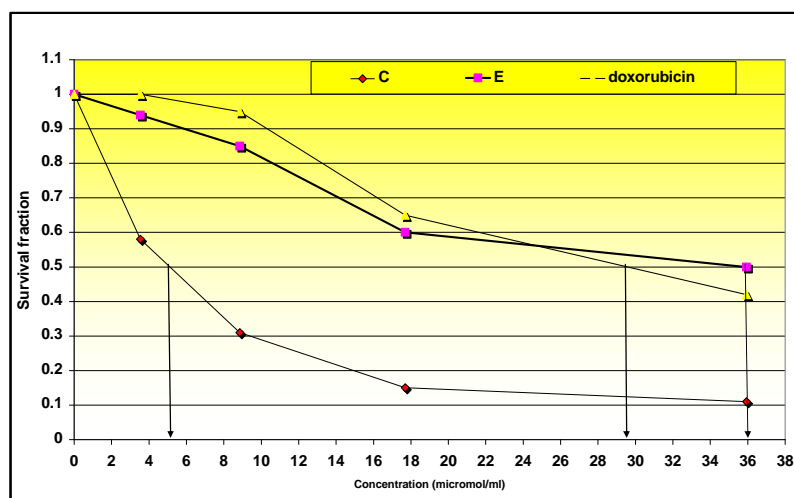
Classic bioisosteres (**7**, **8**), in which the NH group at position 5 is replaced by O or S respectively, showed great reduction in the cytotoxic activity, IC₅₀ = 0.088, 0.354 mM respectively.

3-Chloro-2-(2-pyridylamino)-1,4-naphthoquinone (**10**) was the most cytotoxic compound (IC₅₀ = <0.035 mM) which is higher than that of the standard, doxorubicin (IC₅₀ = 0.15 mM). However, rigidification of this compound (**12**) abolished the antitumor activity. The active compounds (**3**, **10**) were further tested at lower concentrations (0-36.00 μM/mL). The relationship between surviving fraction and drug concentration was plotted to obtain the survival curve of EAC cell line. The response parameter calculated was IC₅₀ value which corresponds to the compound concentration causing 50% mortality in net cells (Table 2, Fig. 1).

Table 2: *In-vitro* cytotoxic activity of some selected synthesized compounds.

Compd. No.	Non Viable cells (%)					IC ₅₀ (mM/mL)
	Concentration (mM mL ⁻¹)					
	0.035	0.088	0.177	0.265	0.354	
Article I doxorubicin	12±1.01	20±1.30	65±2.34	75±4.11	95±2.11	0.15
3	50±1.2-	70±1.9	90±2.9	95±3.1	100±1.0	0.035
4	- ^a	- ^a	- ^a	10	20	>0.354
5	- ^a	2±	5±	10±	10±	>0.354
6	55±3.5	60±3.2	67±4.1	75±2.5	84±2.1	<0.035
7	40±1.4	50±2.1	70±2.3	77±3.0	100±1.01	0.088
8	- ^a	- ^a	2	5	5	>0.354
Article II 10	89±4.8	90±4.3	95±2.1	95±2.9	100±1.0	<0.035
11	30±2.2	11±1.3	30±3.2	44±3.4	50±4.1	0.354
12	- ^a	- ^a	- ^a	- ^a	- ^a	-
13	3±0.01	4±0.011	7±0.012	11.7±0.011	18±0.09	>0.354

^a All cells are alive.



E: compound No. 3; C: compound No. 10.

Fig. 1: Cytotoxic activity of compounds 3, 10 and doxorubicin on cell survival of Ehrlich Ascites Carcinoma cell line (EAC) arrows represented IC₅₀ concentration.

Activity against human cell lines

Antitumor screening was performed at the National Cancer Institute, Cancer Biology Department, and Cairo, Egypt. Potential cytotoxicity of the active compounds **3** and **10** was tested using the method of Skehan *et al*¹⁵. Five human cell lines: liver carcinoma (HEPG2), brain tumor (U251), cervix carcinoma (HELA), breast carcinoma (MCF7) and lung carcinoma (H450) were incubated with five concentrations (0-36.00) $\mu\text{M}/\text{ml}$ ^{15&18} for each compound 5,12-dihydrobenzo[*b*]phenazine-6,11-dione (**3**) and 3-Chloro-2-(2-pyridylamino)-1,4-naphthoquinone

(**10**) showed selective cytotoxicity against the human lung carcinoma cell line (H460) superior to doxorubicin. IC₅₀ of compound **3** (16.25 μM) was 1.3 times higher than that of doxorubicin (IC₅₀ = 20.10 μM)¹⁹. Whereas, IC₅₀ value of compound **10** was 9.90 μM which was 2 times higher than that (20.10 μM) of doxorubicin. These compounds were inactive against liver carcinoma (HEPG2), brain tumor (U251), cervix carcinoma (HELA) and breast carcinoma (MCF7) cell lines. The growth inhibitory action of the selected compounds is summarized in Table 3 and Figure 2.

Table 3: The cytotoxic activity of compounds **3** and **10** and Doxorubicin on Ehrlich Ascites Carcinoma cell line (EAC) and human lung carcinoma cell line (H460).

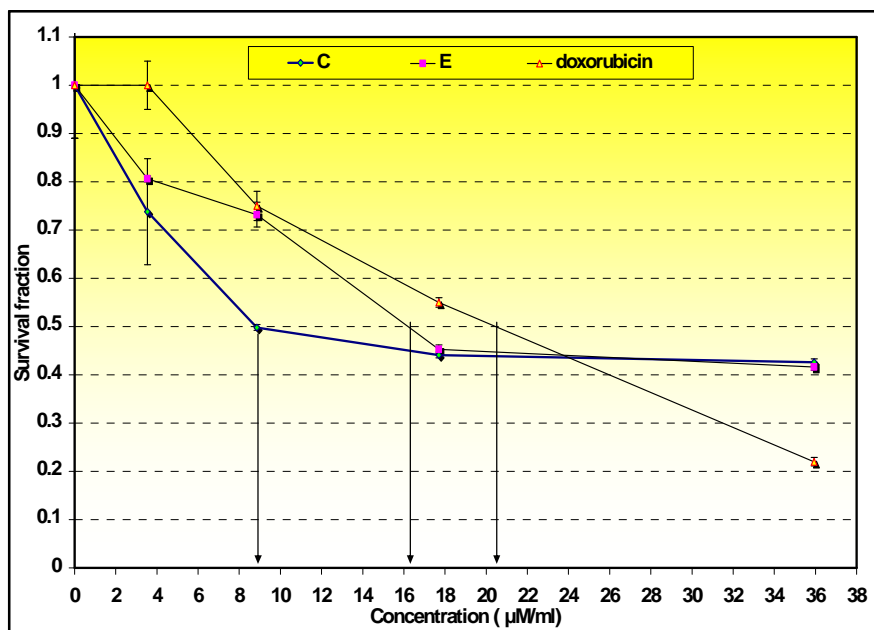
Cpd. No.	Survival fraction					IC ₅₀ ¹ ($\mu\text{M}/\text{mL}$)
	Concentration ($\mu\text{M}/\text{ml}$)					
	0.00	3.54	8.85	17.70	36.00	
Ehrlich ascites Carcinoma (EAC)						
3	1.00±0.11	0.94±0.04	0.85±0.06	0.60±0.02	0.50±0.01	36.00±0.04
10	1.00±0.11	0.58±0.01	0.31±0.03	0.15±0.02	0.11±0.04	5.00±0.10
Ref ²	1.00±0.11	1.00±0.09	0.95±0.05	0.65±0.03	0.42±0.01	29.50±0.06
Lung carcinoma cell line (H460)						
3	1.00±0.11	0.81±0.01	0.73±0.03	0.45±0.01	0.42±0.00	16.25±0.02
10	1.00±0.11	0.74±0.11	0.50±0.01	0.44±0.01	0.43±0.01	9.90±0.07
Ref ²	1.00±0.11	1.00±0.05	0.75±0.03	0.55±0.01	0.22±0.00	20.10±0.06

Results represented as mean of three repeated experiments \pm SE

¹IC₅₀ = Dose of compound which reduces survival of tumor cell line to 50%

The tested compounds were inactive against liver carcinoma (HEPG2), brain tumor (U251), cervix carcinoma (HELA) and breast carcinoma (MCF7) cell line

²Ref = Doxorubicin.



E: compound No. 3; C: compound No. 10.

Fig. 2: Cytotoxic activity of compounds **3** and **10** and doxorubicin on cell survival of lung carcinoma cell line (H460) arrows represented IC50 concentration.

From the study of cytotoxic activity of the tested compounds, it seems that, the introduction of an electron-withdrawing group in position 2 of coplanar tetracyclic compounds, benzo[*b*]phenazine derivatives is favored for activity than electron-donating group. Moreover, the parent compound, 5,12-dihydrobenzo[*b*]phenazine-4,11-dione was ~1.3-fold higher cytotoxic than doxorubicin, however, its classical bioisosteres, 12H-benzo[*b*]phenoxazine and 12H-benzo[*b*]phenothiazine

derivatives were considerably less active. This suggests that, the number of nitrogens is considerably important as well as the presence of electron-withdrawing group which may enable hydrogen bonding with DNA.

3-Chloro-2-(2-pyridylamino)-1,4-naphthoquinone (**10**) having an additional nitrogen showed ~2-fold the cytotoxicity of the standard drug. Rigidification of this compounds (**12,13**) abolished the activity.

REFERENCES

- 1- I. A. Shaikh, F. Johnson and A. P. Grollman, *J. Med. Chem.*, 29, 1329 (1986).
- 2- K. V. Rao and C. P. Rock, *J. Heterocycl. Chem.*, 33, 447 (1996).
- 3- K. V. Rao, *Cancer Chemother. Rep.*, Part 2, 4, 11 (1974).
- 4- A. Lin, B. J. Lillis, *J. Med. Chem.*, 18, 917 (1975)
- 5- W. A. Denny and B. C. Bagulet, In *Molecular Aspects of Anticancer Drug-DNA Interactions*, 2nd ed., Waring, M. J., Neidle, S., Eds., Macmillan: London, 1994, pp. 270-311.
- 6- Y. Pommier, G. Capranico, A. Orr and K. W. Kohn, *Nucleic Acids Rep.*, 19, 5973 (1991).
- 7- P. Fosse, B. Rene, M. C. Le Bret, C. Paoletti and J. M. Saucier, *Nucleic Acids Res.*, 19, 2861 (1991).
- 8- J. C. Wang, *Annu. Rev. Biochem.*, 65, 635 (1996).
- 9- G. W. Rewcastle, W. A. Denny and B. C. Baguley, *J. Med. Chem.*, 40, 1919 (1997).
- 10- Y-S. Kim, S.-Y. Park, H.-J. Lee, M.-E. Suh, D. Schollmeyer and C.-O. Lee, *Bioorg. Med. Chem.*, 11, 1709 (2003).
- 11- N. L. Agarwal and W. Schafer, *J. Org. Chem.*, 45, 5144 (1980).
- 12- I. B. Illescas, N. Martin, J. L. Segura and C. Seoane, *J. Org. Chem.*, 60, 5643 (1995).
- 13- K. V. Rao and W. P. Cullen, *Antibiot. Ann.*, 950 (1959).
- 14- D. J. Brusick, *Cytogenetic Assays, Aberrations and SCE techniques in Carcinogenesis and Mutagenesis testing*, (J.F. Douglas Ed), Human Press Inc.: Clifton, NJ., 1984, pp. 256-76.
- 15- P. Skehan, R. Storeng, D. Scudiero, A. Monks, J. McMahon, D. Vistica, J. T. Warren, H. Bokesch, S. Kenney and M. R. Boyd, *J. Natl. Cancer Inst.*, 82, 1107 (1990).
- 16- G. W. Snedecor and W. G. Cochran, *Statistical Methods' Eight Edition*, Louis State University press, Ames, Iowa, U. (1989).
- 17- J. Y. Lin and Y. C. Chen, *Toxicology.*, 16, 120 (1978).
- 18- P. M. Lorusso, R. Parchment, L. Demchik, J. Knight, L. Polin, J. Dzubow, C. Behrens, B. Harrison, G. Trainor and T. H. Corbett, *Investigational New Drugs*, 16, 287 (1998).
- 19- D. Peer, Y. Dekel, D. Melikhov and R. Margalit, *Cancer Research*, 64, 15, 7562 (2004).