

CHEMICAL CONSTITUENTS OF *GLADIOLUS SEGETUM* KER-GAWL

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من الجزء المذاب في الكلوروفورم لخالصة الكحول الميثيلي لكورمات الجلادبولس سيجيتيم كبير. جول تم فصل تسعة مركبات وقد تم التعرف عليها كالاتي: ثلاثة مركبات من اللجنانات: (+) ديميثوكسي بينوريسينول () (+) بينوريسينول () (+) بينوريسينول أحادي ميثيل إيثر () . مركب نيولجنان: () ديهيدروداي كونيفيريل كحول () . ثلاثة مركبات أنثراكينونية: دي أوكسي إرثرولاسين () فيسيون () ، حمض لاسايك دي ميثيل إستر () بالإضافة الى 6'-O-بلميتويل - 3-O-بيتا سيتوستيرول جلوكوبيرانوزيد () وبيتا سيتوستيرول - 3-O-بيتا جلوكوبيرانوزيد () . وقد تم التعرف على المركبات المفصولة بالوسائل الطبيعية والكروماتوجرافية والطرق الطيفية المختلفة والتي تشتمل على الرنين النووي المغناطيسي ومطياف الكتلة. المركبات من - - تفصل لأول مرة من جنس الجلادبولس بينما المركب رقم () فقد تم فصله سابقا من نفس النبات.

From the chloroform soluble fraction of the methanolic extract of the corms of *Gladiolus segetum* Ker-Gawl (Iridaceae), nine compounds were isolated and identified as follows: the lignans (+)-demethoxypinoresinol (1), (+)-pinoresinol (2) and (+)-pinoresinol monomethylether (3); the neolignan (-)-dehydrodiconiferyl alcohol (4) and the anthraquinones deoxyerythrolaccin (5), physcion (6) and laccaic acid D methylester (7) together with 6'-O-palmitoyl-3-O- β -sitosterol glucoside (8) and β -sitosterol-3-O-glucoside (9). The structures of the isolated compounds were determined by physical and spectroscopic methods including NMR and MS spectral analysis. Compounds 1-4 and 6-9 are reported here for the first time from the genus *Gladiolus* while compound 5 was previously isolated from the same plant.

INTRODUCTION

Gladiolus segetum Ker-Gawl (F. Iridaceae) is a herbaceous plant cultivated in Egypt as an ornamental plant.^{1,2} In the folk medicine, the uses of some *Gladiolus* species were reported as a remedy for dysentery, impotence and for relief of rheumatic pains. Moreover, the smoke from the burning corms is sometimes inhaled for colds.³ The current literatures revealed that the genus *Gladiolus* showed the presence of flavonols, anthocyanidins, ascorbic acid, saponins, fatty acids, mucilage and anthraquinones.⁴⁻¹¹

The present study deals with the isolation and structure elucidation of nine compounds from the chloroform soluble fraction of the methanolic extract of the corms of *G. segetum* using different chromatographic techniques and various tools of NMR spectral analysis

including H-H COSY and C-H COSY as well as EI MS spectral analysis. The isolated compounds (1-9) were identified as the lignans (+)-demethoxypinoresinol (1), (+)-pinoresinol (2) and (+)-pinoresinol monomethylether (3); the neolignan (-)-dehydrodiconiferyl alcohol (4) and the anthraquinones deoxyerythrolaccin (5), physcion (6) and laccaic acid D methylester (7) together with 6'-O-palmitoyl-3-O- β -sitosterol glucoside (8) and β -sitosterol-3-O-glucoside (9).

EXPERIMENTAL

Melting points: uncorrected and measured on Stuart Scientific (SMPI). Optical rotation was measured on Union PM-101 automatic digital polarimeter. Nuclear Magnetic Resonance (400 MHz for ¹H NMR and 100 MHz for ¹³C NMR) spectra were recorded on a

JEOL JNM α -400 spectrometer using TMS as internal standard. Mass spectra were taken on a JEOL JMS-SX 102 spectrometer by direct inlet method at an ionizing voltage of 70 eV. For column chromatography, Kieselgel 60 (70-230 mesh, Merck) and Sephadex LH-20 (Merck) were used. For thin layer chromatography, silica gel 60 precoated plates, F-254 (Merck) were used with the following solvent systems:

I- CHCl_3 - MeOH (9.5 : 0.5)

II- CHCl_3 - MeOH (9 : 1)

III- CHCl_3 - MeOH (8 : 2)

Plant material

Corms of *Gladiolus segetum* Ker-Gawl were collected from the plants cultivated in the Experimental Station of Pharmacognosy Department, Faculty of Pharmacy, Al-Azhar University, Assiut Branch, Assiut, Egypt in 1999. The plant was identified by Dr. Salah El-Nagar, Professor of Taxonomy, Dept. of Botany, Faculty of Science, Assiut University.

Extraction and isolation

The air-dried powdered corms of *Gladiolus segetum* (500 g) was extracted three times by maceration with MeOH at room temperature. The methanolic extracts were combined together and concentrated under reduced pressure till dryness. The dried methanolic extract (145 g) was suspended in H_2O and fractionated with *n*-hexane and CHCl_3 successively. The CHCl_3 fraction (6 g) was chromatographed on silica gel column and eluted with CHCl_3 and CHCl_3 with increasing gradient of MeOH to give 12 fractions (F-1 to F-12).

Fraction F-3 (27 mg) was chromatographed on Sephadex LH-20 column using MeOH as a solvent system to afford compound **6** (19 mg). Fraction F-4 (127 mg) was chromatographed on silica gel column using CHCl_3 - MeOH (9.7 : 0.3) as a solvent system to give compounds **3** (20 mg) and **4** (28 mg). Fraction F-5 (58 mg) was subjected to preparative TLC using silica gel plates and CHCl_3 - MeOH (9.5 : 0.5) as a solvent system to afford compound **2** (37 mg). Fraction F-6 (69 mg) was subjected to Sephadex LH-20 column chromatography using MeOH as a solvent system to give compounds **1** (12 mg) and **7** (23 mg). Fraction F-7 (26 mg) was chromatographed on Sephadex LH-20 column using

MeOH as a solvent system to afford compound **5** (10 mg). Fraction F-8 (168 mg) was applied on silica gel column chromatography using CHCl_3 - MeOH (9 : 1) as a solvent system to give compound **8** (57 mg). Fraction F-9 (563 mg) was crystallized using MeOH to afford compound **9** (394 mg). Fractions F-1 and F-2 were resinous in nature and impure while the quantity of fractions F-10, F-11 and F-12 was very minor.

Compound (1)

Amorphous powder, $R_f = 0.42$ (System I), $[\alpha]_D^{26} +88.2^\circ$ (CHCl_3 , 0.04). EI-MS (m/z): 328 (M^+ , $\text{C}_{19}\text{H}_{20}\text{O}_5$). ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 7.16 (2H, *d*, $J = 8.5$ Hz, H-2, 6), 7.14 (1H, *d*, $J = 8.0$ Hz, H-5), 6.87 (1H, *d*, $J = 1.8$ Hz, H-2), 6.74 (1H, *dd*, $J = 8.0, 1.8$ Hz, H-6), 6.71 (2H, *d*, $J = 8.5$ Hz, H-3, 5), 4.62 and 4.60 (each 1H, *d*, $J = 5.1$ Hz, H-2, 6), 4.10 and 3.71 (each 2H, *m*, H-4, 8), 3.74 (3H, *s*, -OMe), 3.03 (2H, *m*, H-1, 5). ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$, Table 1).

Compound (2)

Colourless crystals (MeOH), m.p 117-119 $^\circ$, $R_f = 0.51$ (System I), $[\alpha]_D^{26} +76.2^\circ$ (CHCl_3 , 0.07) lit.¹² $+77.5^\circ$. EI-MS (m/z): 358 (M^+ , $\text{C}_{20}\text{H}_{22}\text{O}_6$). ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 6.71-6.83 (6H, *m*, aromatic H), 4.57 (2H, *d*, $J = 4.12$ Hz, H-2, 6), 4.08 and 3.73 (each 2H, *m*, H-4, 8), 3.78 (6H, *s*, 2x-OMe), 3.04 (2H, *m*, H-1, 5). ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$, Table 1).

Compound (3)

Amorphous powder, $R_f = 0.54$ (System I), $[\alpha]_D^{26} +64.2^\circ$ (CHCl_3 , 1.2) lit.¹³ $+65.6^\circ$. EI-MS (m/z): 372 (M^+ , $\text{C}_{21}\text{H}_{24}\text{O}_6$). ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 6.69-6.85 (6H, *m*, aromatic H), 4.65 (2H, *d*, $J = 4.8$ Hz, H-2, 6), 4.27 and 3.77 (each 2H, *m*, H-4, 8), 3.73 (3H, *s*, -OMe), 3.72 (6H, *s*, 2x-OMe), 3.08 (2H, *m*, H-1, 5). ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$, Table 1).

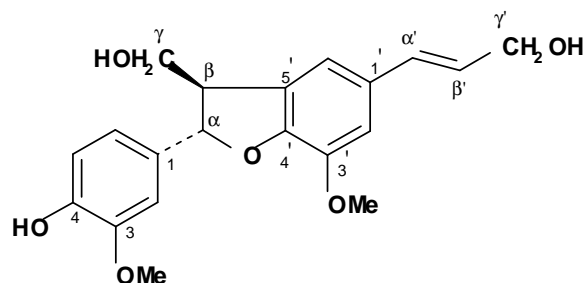
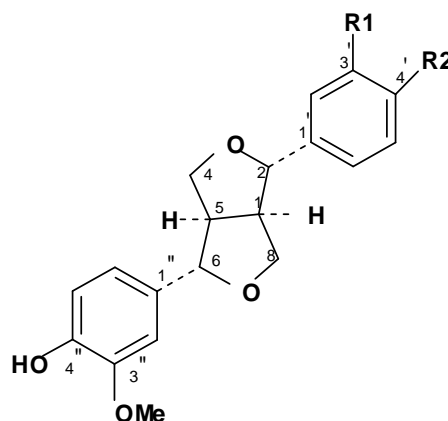
Compound (4)

Amorphous powder, $R_f = 0.56$ (System I), $[\alpha]_D^{26} -31.3^\circ$ (CHCl_3 , 1.15) lit.¹⁴ -31.6° . EI-MS (m/z): 358 (M^+ , $\text{C}_{20}\text{H}_{22}\text{O}_6$). ^1H NMR (400 MHz, $\text{DMSO}-d_6$): δ 6.73-6.89 (5H, *m*, aromatic H), 6.45 (1H, *d*, $J = 16.1$ Hz, H- α), 6.17 (1H, *m*, H- β), 5.44 (1H, *d*, $J = 6.4$ Hz, H- α), 4.07 (2H, *br.d*, $J = 5.1$ Hz, H- γ), 3.84 (2H, *m*, H- γ),

Compound (1): R1= H, R2= OH

Compound (2): R1= OMe, R2= OH

Compound (3): R1, R2= OMe

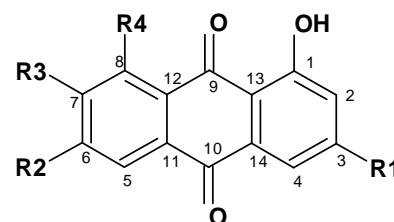


Compound (4)

Compound (5): R1, R2= OH, R3= H, R4= Me

Compound (6): R1= Me, R2= OMe, R3= H, R4= OH

Compound (7): R1, R2= OH, R3= COOMe, R4= Me



3.77 (3H, *s*, -OMe), 3.75 (3H, *s*, -OMe), 3.41 (1H, *m*, H- β). ^{13}C NMR (100 MHz, DMSO- d_6 , Table 1).

Compound (5)

Orange crystals from MeOH, m.p 300° (decomp.), R_f = 0.45 (System II). EI-MS (m/z): 270 (M^+ , $\text{C}_{15}\text{H}_{10}\text{O}_5$). ^1H NMR (400 MHz, DMSO- d_6): δ 13.24 (1H, *s*, -OH at C-1), 7.42 (1H, *d*, J = 2.5 Hz, H-5), 7.04 (1H, *d*, J = 2.3 Hz, H-4), 6.99 (1H, *d*, J = 2.5 Hz, H-7), 6.55 (1H, *d*, J = 2.3 Hz, H-2), 2.70 (3H, *s*, -Me). ^{13}C NMR (100 MHz, DMSO- d_6 , Table 2).

Compound (6)

Orange yellow crystals from MeOH, m.p 235-236°, R_f = 0.71 (System I). EI-MS (m/z):

284 (M^+ , $\text{C}_{16}\text{H}_{12}\text{O}_5$). ^1H NMR (400 MHz, DMSO- d_6): δ 12.41 and 12.18 (each 1H, *s*, -OH at C-1 and C-8), 7.36 (1H, *d*, J = 2.0 Hz, H-4), 7.08 (1H, *d*, J = 1.8 Hz, H-5), 6.94 (1H, *d*, J = 2.0 Hz, H-2), 6.78 (1H, *d*, J = 1.8 Hz, H-7), 3.88 (3H, *s*, -OMe), 2.66 (3H, *s*, -Me). ^{13}C NMR (100 MHz, DMSO- d_6 , Table 2).

Compound (7)

Orange crystals from MeOH, m.p 272-273°, R_f = 0.36 (System I). EI-MS (m/z): 328 (M^+ , $\text{C}_{17}\text{H}_{12}\text{O}_7$). ^1H NMR (400 MHz, DMSO- d_6): δ 13.30 (1H, *s*, -OH at C-1), 7.65 (1H, *s*, H-5), 7.13 (1H, *d*, J = 2.1 Hz, H-4), 6.74 (1H, *d*, J = 2.1 Hz, H-2), 3.60 (3H, *s*, -COOMe), 2.65 (3H, *s*, -Me). ^{13}C NMR (100 MHz, DMSO- d_6 , Table 2).

Table 1: ^{13}C NMR spectral data of compounds **1-4** (100 MHz, DMSO- d_6).

C	1	2	3	C	4
1	53.6 ^a <i>d</i>	53.9 <i>d</i>	53.6 <i>d</i>	1	132.4 <i>s</i>
2	85.2 ^b <i>d</i>	85.6 <i>d</i>	84.5 <i>d</i>	2	110.3 <i>d</i>
4	70.9 ^c <i>t</i>	71.3 <i>t</i>	71.1 <i>t</i>	3	147.6 <i>s</i>
5	53.5 ^a <i>d</i>	53.9 <i>d</i>	53.6 <i>d</i>	4	146.4 <i>s</i>
6	85.0 ^b <i>d</i>	85.6 <i>d</i>	84.5 <i>d</i>	5	115.3 <i>d</i>
8	70.8 ^c <i>t</i>	71.3 <i>t</i>	71.1 <i>t</i>	6	118.6 <i>d</i>
1	131.6 <i>s</i>	132.6 <i>s</i>	132.1 <i>s</i>	1	129.5 <i>s</i>
2	127.6 <i>d</i>	110.0 <i>d</i>	110.0 <i>d</i>	2	110.4 <i>d</i>
3	115.1 <i>d</i>	148.2 <i>s</i>	146.2 <i>s</i>	3	143.7 <i>s</i>
4	156.7 <i>s</i>	146.5 <i>s</i>	144.9 <i>s</i>	4	147.2 <i>s</i>
5	115.1 <i>d</i>	115.5 <i>d</i>	114.0 <i>d</i>	5	130.6 <i>s</i>
6	127.6 <i>d</i>	119.0 <i>d</i>	118.2 <i>d</i>	6	115.0 <i>d</i>
1	132.2 <i>s</i>	132.6 <i>s</i>	133.3 <i>s</i>	α	87.2 <i>d</i>
2	110.3 <i>d</i>	110.0 <i>d</i>	108.4 <i>d</i>	β	53.0 <i>d</i>
3	147.5 <i>s</i>	148.2 <i>s</i>	148.7 <i>s</i>	γ	62.9 <i>t</i>
4	145.9 <i>s</i>	146.5 <i>s</i>	148.1 <i>s</i>	α	129.0 <i>d</i>
5	115.0 <i>d</i>	115.5 <i>d</i>	110.8 <i>d</i>	β	128.0 <i>d</i>
6	118.6 <i>d</i>	119.0 <i>d</i>	117.9 <i>d</i>	γ	61.7 <i>t</i>
<u>OMe</u>	55.6 <i>q</i>	56.1 <i>q</i> (2xOMe)	55.5 <i>q</i> (2xOMe) 55.6 <i>q</i>	<u>OMe</u>	55.5 <i>q</i> 55.6 <i>q</i>

^{a, b, c} values may be interchangeable within each column

Table 2: ^{13}C NMR spectral data of compounds **5-7** (100 MHz, DMSO- d_6).

C	5	6	7
1	164.2 <i>s</i>	164.4 <i>s</i>	160.0 <i>s</i>
2	108.2 <i>d</i>	105.8 <i>d</i>	107.9 <i>d</i>
3	164.5 <i>s</i>	145.1 <i>s</i>	157.1 <i>s</i>
4	112.1 <i>d</i>	121.2 <i>d</i>	107.0 <i>d</i>
5	107.1 <i>d</i>	125.1 <i>d</i>	112.7 <i>d</i>
6	161.7 <i>s</i>	162.7 <i>s</i>	156.4 <i>s</i>
7	124.5 <i>d</i>	106.8 <i>d</i>	130.0 <i>s</i>
8	144.8 <i>s</i>	164.7 <i>s</i>	140.4 <i>s</i>
9	187.9 <i>s</i>	187.8 <i>s</i>	188.3 <i>s</i>
10	182.3 <i>s</i>	182.5 <i>s</i>	181.7 <i>s</i>
11	134.2 <i>s</i>	134.1 <i>s</i>	132.8 <i>s</i>
12	122.3 <i>s</i>	112.9 <i>s</i>	121.8 <i>s</i>
13	109.9 <i>s</i>	136.8 <i>s</i>	111.3 <i>s</i>
14	136.6 <i>s</i>	110.9 <i>s</i>	136.8 <i>s</i>
Me	22.3 <i>q</i>	23.6 <i>q</i>	20.1 <i>q</i>
OMe		56.1 <i>q</i>	
<u>COOMe</u>			167.8 <i>s</i>
<u>COOMe</u>			52.6 <i>q</i>

Compound (8)

Yellow greasy substance, $R_f = 0.65$ (System III). ^1H NMR (400 MHz, CDCl_3): δ 5.22 (1H, *m*, H-6), 5.03 (1H, *d*, $J = 7.8$ Hz, H-1 anomeric proton of the glucopyranosyl moiety), 4.05-4.32 (5H, *m*, sugar protons H-2 to H-6), 3.57 (1H, *m*, H-3), 0.68-1.09 (7xMe). ^{13}C NMR (100 MHz, CDCl_3 , Table 3).

Table 3: ^{13}C NMR spectral data of compound **8** (100 MHz, CDCl_3).

β -sitosterol moiety				Glucopyranosyl moiety	
C		C		C	
1	36.9	16	26.9	1	101.1
2	29.6	17	55.6	2	73.4
3	78.7	18	11.7	3	76.4
4	39.1	19	18.0	4	70.3
5	139.2	20	35.1	5	76.7
6	119.1	21	18.7	6	63.6
7	31.6	22	34.0	Palmitoyl moiety	
8	31.5	23	25.3	C	
9	49.1	24	45.1	1	173.5
10	36.0	25	28.9	2	34.0
11	20.8	26	18.6	3 -15	22.3-29.4
12	39.3	27	19.0	16	13.8
13	40.6	28	22.4		
14	55.8	29	11.9		
15	23.5				

Compound (9)

Amorphous powder, $R_f = 0.56$ (System III), identified as β -sitosterol-3-O- β -glucopyranoside by TLC co-chromatography with authentic sample.

RESULTS AND DISCUSSION

The air-dried powdered corms of *G. segetum* was extracted with methanol and the extract was fractionated with *n*-hexane and chloroform successively. From the chloroform soluble fraction, nine compounds (1-9) were isolated.

The ^{13}C NMR including DEPT mode measurements (Table 1) and ^1H NMR spectral data of compound **1** showed characteristic features of tetrahydrofuran lignans and suggesting that it belongs to the 2,6-diaryl-3,7-dioxabicyclo{3.3.0}octane type lignan.^{15,16} The EI-MS spectral analysis of **1** exhibited a

molecular ion peak at m/z 328 (M^+) corresponding to a molecular formula $\text{C}_{19}\text{H}_{20}\text{O}_5$ and intense fragments at m/z 93 and 123 for 4-hydroxyphenyl and guaiacyl moieties, respectively. The presence of 4-hydroxyphenyl group was assigned from the ^{13}C NMR spectral data (Table 1) at δ_{C} 131.6 *s* (C-1), 127.6 *d* (2C, C-2 and 6), 115.1 *d* (2C, C-3 and 5) and 156.7 *s* (C-4) and ^1H NMR signals at δ_{H} 7.16 (2H, *d*, $J = 8.5$ Hz, H-2, 6) and 6.71 (2H, *d*, $J = 8.5$ Hz, H-3, 5). The signals at δ_{C} 132.2 *s* (C-1), 110.3 *d* (C-2), 147.5 *s* (C-3), 145.9 *s* (C-4), 115.0 *d* (C-5), 118.6 *d* (C-6) and 55.6 *q* (OMe) with the ^1H NMR signals (ABX system) at δ 6.87 (1H, *d*, $J = 1.8$ Hz, H-2), 7.14 (1H, *d*, $J = 8.0$ Hz, H-5) and 6.74 (1H, *dd*, $J = 8.0, 1.8$ Hz, H-6) in addition to the aromatic methoxyl group at 3.74 (3H, *s*) indicated the presence of 4-hydroxy-3-methoxyphenyl group (guaiacyl moiety). The presence of 3,7-dioxabicyclo{3.3.0}octane was indicated from the NMR data at δ_{C} 53.6 *d*, 53.5 *d* (C-1 and 5); 85.2 *d*, 85.0 *d* (C-2 and 6) and 70.9 *t*, 70.8 *t* (C-4, 8) with δ_{H} 3.03 (2H, *m*, H-1, 5), 4.62 and 4.60 (each 1H, *d*, $J = 5.1$ Hz, H-2, 6) and 4.10 and 3.71 (each 2H, *m*, H-4, 8). The above mentioned assignments were confirmed by measurements of C-H COSY and H-H COSY.

Concerning the stereochemistry of compound **1**, it is known that the fusion of the tetrahydrofurans in this class of lignans to obtain the bicycle is always *Cis*.¹⁶ The ^1H NMR signals of the benzylic protons H-2 and H-6 at δ_{H} 4.62 and 4.60 showed that both possessed a diequatorial stereochemistry since they appeared between δ_{H} 3.75 and 4.70 whereas in the diaxial series they appear between δ_{H} 3.25 and 4.0 and the other ^1H NMR signals fit well for the stereochemistry.¹⁶

The above mentioned MS and NMR results of compound **1** are in a good agreement with those reported in the literature for (1*R**, 2*S**, 5*R**, 6*S**)-2-(4-hydroxyphenyl)-6-(3-methoxy-4-hydroxyphenyl)-3,7-dioxabicyclo{3.3.0}octane which is a demethoxylated derivative of pinoresinol¹⁵, due to the positive value of its optical rotation it can be named as (+)-demethoxypinoresinol. This compound was previously isolated from *Mikania saltensis* (Compositae)¹⁵ but due to the scarcity of the isolated material its ^{13}C NMR spectral analysis was not reported, it is herein given (Table 1).

The ^{13}C NMR (Table 1) and ^1H NMR spectral data of compound **2** showed a similarity with those of **1** and indicated that it is a tetrahydrofuran lignan derivative of 2,6-diaryl-3,7-dioxabicyclo{3.3.0}octane type with the presence of two guaiacyl moieties as deduced from the signals at δ_{C} 132.6 *s* (2C, C-1 and 1), 110.0 *d* (2C, C-2 and 2), 148.2 *s* (2C, C-3 and 3), 146.5 *s* (2C, C-4 and 4), 115.5 *d* (2C, C-5 and 5), 119.0 *d* (2C, C-6 and 6) and 56.1 (2xOMe) with δ_{H} 6.71-6.83 (6H, *m*, aromatic H) and 3.78 (6H, *s*, 2xOMe). The NMR signals of the two fused tetrahydrofuranes displayed at δ_{C} 53.9 *d* (2C, C-1 and 5), 85.6 *d* (2C, C-2 and 6) and 71.3 *t* (2C, C-4 and 8) with δ_{H} 3.04 (2H, *m*, H-1, 5), 4.57 (2H, *d*, $J=4.12$ Hz, H-2, 6) and 4.08 and 3.73 (each 2H, *m*, H-4, 8). The EI-MS spectral data of compound **2** supported the above assignments, whereas the molecular ion peak displayed at m/z 358 (M^+) corresponding to the molecular formula $\text{C}_{20}\text{H}_{22}\text{O}_6$ and indicated the presence of an additional methoxyl group than compound **1**. For further confirmation, 2D NMR spectral analysis was carried out including H-H COSY and C-H COSY.

From the above mentioned data, compound **2** was identified as (+)-pinoresinol; (1*R**,2*S**,5*R**,6*S**)-2-(3-methoxy-4-hydroxyphenyl)-6-(3-methoxy-4-hydroxyphenyl)-3,7-dioxabicyclo{3.3.0}octane which was previously isolated from several plant species.^{12,15,17}

The EI-MS spectral analysis of compound **3** exhibited a molecular ion peak at m/z 372 (M^+) coincident with the molecular formula $\text{C}_{21}\text{H}_{24}\text{O}_6$. The ^{13}C NMR (Table 1) and ^1H NMR spectral data of compound **3** were similar to those of compound **2** except for the presence of three aromatic methoxyl groups at δ_{C} 55.9 (2xOMe) and 55.6 (OMe) with δ_{H} 3.72 (6H, *s*, 2xOMe) and 3.73 (3H, *s*, OMe) instead of two methoxyls in compound **2** which indicated that the two aryl moieties of the lignan derivative **3** are guaiacyl and 4-O-methylguaiacyl moieties.

The above mentioned data together with the measurements of H-H COSY and C-H COSY confirmed the identity of compound **3** as (+)-pinoresinol monomethylether; (1*R**,2*S**,5*R**,6*S**)-2-(3,4-dimethoxyphenyl)-6-(3-methoxy-4-hydroxyphenyl)-3,7-dioxabicyclo{3.3.0}octane which was previously isolated from *Forsythia japonica* and *F. giraldiana*.¹³

The EI-MS spectral analysis of compound **4** showed a molecular ion peak at m/z 358 (M^+) assignable to a molecular formula $\text{C}_{20}\text{H}_{22}\text{O}_6$. The ^{13}C NMR including DEPT mode measurements (Table 1) and ^1H NMR spectral data of compound **4** showed a similar pattern to those reported for dihydrobenzofuran type neolignans.^{14,18} The presence of α,β -dihydrobenzofuran could be recognized from the NMR spectral data at δ_{H} 5.44 (1H, *d*, $J=6.4$ Hz) and 3.41 (1H, *m*) assignable to the methine protons H- α and H- β , respectively with δ_{C} 87.2 *d* (C- α) and 53.0 *d* (C- β) of the dihydrofuran ring fused to a coniferyl alcohol moiety to form the benzofuran skeleton. The *trans* 3-hydroxyprop-2-enyl group of the coniferyl alcohol at C-1 was deduced from the signals at δ_{H} 6.45 (1H, *d*, $J=16.1$ Hz, H- α) and 6.17 (1H, *m*, H- β) for the *trans* olefinic protons, and at 4.07 (2H, *br.d*, $J=5.1$ Hz) for the hydroxylated methylene H- γ with δ_{C} 129.0 *d* (C- α), 128.0 *d* (C- β) and 61.7 (C- γ). The C-2 and 6 of the coniferyl alcohol moiety were displayed at δ_{C} 110.4 *d* and 115.4 *d*, respectively, and the other aromatic carbons found at δ_{C} 129.5 *s*, 143.7 *s*, 147.2 *s* and 130.6 *s* were assigned to C-1, 3, 4 and 5, respectively. The presence of a hydroxymethyl group at C- β of the dihydrofuran ring was deduced from the ^1H NMR signal at δ 3.84 (2H, *m*, H- γ) with δ_{C} 62.9. Moreover, the ^{13}C NMR spectral data revealed also the presence of a guaiacyl moiety linked to C- α of the dihydrofuran ring from the signals at δ_{C} 132.4 *s* (C-1), 110.3 *d* (C-2), 147.6 *s* (C-3), 146.4 *s* (C-4), 115.3 *d* (C-5), 118.6 *d* (C-6) as well as the methoxyl group at C-3 displayed at either δ_{C} 55.5 or 55.6 (interfered with that of the coniferyl alcohol moiety at C-3). This suggestion was confirmed from the ^1H NMR data at δ 6.73-6.89 (*m*) of the aromatic protons H-2, 5 and 6 (interfered with those of H-2 and 6).

The measurement of H-H COSY and C-H COSY together with comparing the above mentioned spectral data with those reported in the literature,¹⁴ as well as, the result of the optical rotation analysis of compound **4** (Experimental section) confirmed its identity as the neolignan (-)-dehydroconiferyl alcohol which was previously reported from *Cistanche tubulosa*.¹⁴

The EI-MS, ^1H NMR, ^{13}C NMR (Table 2), H-H COSY and C-H COSY as well as comparing the physical properties and spectral data with those reported earlier^{8,19,20} indicated that compound **5** is the anthraquinone deoxyerythrolaccin which was previously isolated from the same plant by Abdallah *et al.*,⁸ however the numbering system mentioned in this reference is not coincident with the general numbering system of anthraquinones.¹⁹⁻²¹ The corrected one according to that reported earlier¹⁹⁻²¹ is herein given.

The EI-MS spectral analysis of compound **6** exhibited a molecular ion peak at m/z 284 (M^+) corresponding to the molecular formula $\text{C}_{16}\text{H}_{12}\text{O}_5$. The ^1H NMR spectral data showed four *meta*-coupled protons at δ_{H} 6.94 and 7.36 (each 1H, *d*, $J = 2.0$ Hz) assignable to H-2 and H-4, respectively, while the signals at δ 7.08 and 6.78 (each 1H, *d*, $J = 1.8$ Hz) corresponding to H-5 and H-7, respectively. It showed also two signals of chelated hydroxyl groups at C-1 and C-8 at δ 12.41 and 12.18 (each 1H, *s*) and signals of an aromatic methyl at δ 2.66 (3H, *s*) and aromatic methoxyl at δ 3.88 (3H, *s*).

Comparison the EI-MS, ^1H NMR, ^{13}C NMR (Table 2) and physical properties of compound **6** with those reported in the literature indicated that it is the anthraquinone physcion.²¹⁻²³

The EI-MS of compound **7** showed a molecular ion peak at m/z 328 (M^+) corresponding to a molecular formula $\text{C}_{17}\text{H}_{12}\text{O}_7$. The ^1H NMR spectral data showed signals at δ 13.30 (1H, *s*) assigned to a chelated hydroxyl group at C-1, 7.65 (1H, *s*) assigned to H-5, a *meta*-coupled protons at δ 6.74 and 7.13 (each 1H, *d*, $J = 2.1$ Hz, H-2 and 4), a signal of ester methyl at C-7 at δ 3.60 (3H, *s*, -COOMe) and an aromatic methyl at 2.65 (3H, *s*, Me at C-8). The ^{13}C NMR spectral analysis including DEPT mode measurements (Table 2) confirmed the presence of a methylester group from the signals at δ_{C} 167.8 *s* and 52.6 *q*, an aromatic methyl at δ 20.1, two carbonyl carbons at δ 188.3 *s* and 181.7 *s* for C-9 and C-10, respectively, and three hydroxylated aromatic carbons at 160.0 *s*, 157.1 *s* and 156.4 *s* corresponding to C-1, 3 and 6, respectively. The measurements of H-H COSY and C-H COSY confirmed these results.

Comparing the above mentioned data with those reported earlier¹⁹ indicated that compound **7** is laccic acid D methylester that

was previously isolated from *Aloe saponaria*.¹⁹ Anthraquinones with similar substitution pattern specially those with carboxyl group at *ortho* position to both methyl and methoxyl or hydroxyl functions were previously isolated from the genus *Gladiolus*.^{8,10,11}

The ^1H NMR and ^{13}C NMR (Table 3) spectral data of compound **8** suggested the presence of β -sitosterol 3-O- β -glucopyranosyl and palmitoyl moieties.²⁴ The attachment of the palmitoyl moiety at C-6 of the glucopyranosyl moiety was assigned from the downfield shift of C-6 to δ_{C} 63.6. Comparison the NMR data of compound **8** with those of authentic sample indicated that it is 6-O-palmitoyl-3-O- β -sitosterol glucopyranoside.²⁴

Compound **9** can be identified as β -sitosterol-3-O- β -glucopyranoside by co-chromatography with an authentic sample.

Compounds **1-4** and **6-9** are reported here for the first time from the genus *Gladiolus* while compound **5** was previously isolated from the same plant. Finally, the occurrence of lignans, neolignans and anthraquinones in the genus *Gladiolus* is noteworthy from the chemotaxonomical viewpoint.

REFERENCES

- 1- J. Hutchinson, "The Families of Flowering Plants", 3rd. Edn., The Clarendon Press, Oxford, 1973, p. 805.
- 2- R. Muschler, "A Manual Flora of Egypt", Verlag Von J. Cramer, New York, 1970, p. 238.
- 3- J. M. Watt, "Medicinal and Poisonous Plants of Southern and Eastern Africa", 2nd. Edn., E & S. Livingstone Ltd., Edinburg and London, 1962, p. 504.
- 4- A. Solehian, Bull. Trav. Soc. Pharm. Lyon, 17 (3), 86 (1973).
- 5- T. Hibert and I. Nahrung, 1, 27 (1957), through C.A. 52: 3924.
- 6- W. Richard and W. Hoehne, Farm. Eodontal. Univ. Sao Paulo, 9, 17 (1951), through C.A. 46: 10548.
- 7- A. A. Ali, M. A. El-Shanawany, M. K. Mesbah and S. A. Ross, Bull. Pharm. Sci. Assiut Univ., 8, 109 (1985).
- 8- A. A. Ali, O. M. Abd-Allah and W. Steglich, Phytochemistry, 28, 281 (1989).

- 9- A. M. H. Nafady, "Phytochemical Studies on *Corchorus olitorius* L. (Tiliaceae) and *Gladiolus segetum* Ker-Gawl (Iridaceae)", A Master Thesis, Faculty of Pharmacy, Assiut University (2000).
- 10- D. Y. Wang, Q. Ye, G. L. Zhang and B. G. Li, *J. Asian Nat. Prod. Res.*, 5 (4), 297 (2003).
- 11- D. Y. Wang, Q. Ye, B. G. Li and G. L. Zhang, *Nat. Prod. Res.*, 17 (5), 365 (2003).
- 12- H. Tsukamoto, S. Hisada and S. Nishibe, *Chem. Pharm. Bull.*, 32 (11), 4482 (1984).
- 13- S. Kitagawa, S. Nishibe, R. Benecke and H. Thieme, *Chem. Pharm. Bull.*, 36 (9), 3667 (1988).
- 14- F. Yoshizawa, T. Deyama, N. Takizawa, K. Usmanhani and M. Ahmad, *Chem. Pharm. Bull.*, 38 (7), 1927 (1990).
- 15- M. D. R. Cuena and C. A. N. Catalan, *J. Nat. Prod.*, 54 (4), 1162 (1991).
- 16- A. C. Casabuono and A. B. Pomilio, *Phytochemistry*, 35 (2), 479 (1994).
- 17- G. M. Massanet, E. Pando, F. R. Luis and E. Zubia, *Fitoterapia*, Vol. LX (1), 3 (1989).
- 18- H. Achenbach, J. Groß, X. A. Dominguez, G. Cano, J. V. Star, L. D. C. Brussolo, G. Munoz, F. Salgado and L. Lopez, *Phytochemistry*, 26 (4), 1159 (1987).
- 19- A. Yagi, K. Makino and I. Nishioka, *Chem. Pharm. Bull.*, 22 (5), 1159 (1974).
- 20- H. Abd El-Fattah, *International Journal of Pharmacognosy*, 35 (4), 1 (1997).
- 21- R. H. Thomson, "Naturally Occurring Quinones", 2nd Edn., Academic Press, London and New York, (1971).
- 22- N. B. Mulchandani and S. A. Hassarajani, *Planta Medica*, 32, 357 (1977).
- 23- E. Y. Bakheet, "Phytochemical Study of *Cassia didymobotrya* Fres. (F. Leguminosae) Growing in Egypt", A Ph. D. Thesis, Faculty of Pharmacy, Assiut University (1989).
- 24- S. M. El-Sayyad, M. H. Mohamed, M. S. Kamel, K. M. Mohamed and A. N. El-Hifnawy, *Bull. Pharm. Sci. Assiut Univ.*, 22 (2), 123 (1999).