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NEW REAGENTS FOR DETECTION OF PHENOTHIAZINE DRUGS ON THIN LAYER PLATES

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ABSTRACT

About 50 new regeants have been developed for the detection of phenothiazine drugs on TLC plates by giving different colours. Each reagent consists of an amine and an oxidant. Most of the developed reagents are very sensitive for detection of even lug phenothiazine drug.

INTRODUCTION

Many reagents are available for the detection of phenothiazine drugs on paper and thin layer chromatograms. Among these reagents are sulphuric acid, Marquis' reagents, Dragendorff's reagent, ceric sulphate, low hydrogen peroxide, 5% potassium permanganate, 2% ferric chloride, and diazotized p-nitroaniline.

New reagents for Astection of phenothiasine drugs on thin layer plates

The purpose of this investigation was to develop new reagents for the detection of different phenothiazine drugs. All these reagents consisted of an amine and an oxidant prepared separately. The amine may be primary or secondary (aliphatic or aromatic). The oxidants used are N-bromosuccinimide(NBS), ceric sulphate, chlorine water and potassium periodate.

EXPERIMENTAL

Materials and Reagents

Pure drug samples examined were of pharmaceutical grade, obtained as gifts from various manufacturers, and were utilized as working reference compounds without further treatment. All solvents were of analytical grade.

Precoated silica gel plates (0.25 mm) without fluorescence (Merck, Darmstadt, FRG) were used without prior activation.

Oxidant solutions: 3% w/v aqueous solution of NBS, ceric sulphate and potassium periodate. Saturated solution of chlorine in distilled water.

Amine solutions-3% v/v or w/v Methanolic solution of each amine stated in the Tables.

Developing solvents⁹: Ammonium acetate: water: methanol (3:20:100, w/v/v).

Phenothiazine Drug Solutions:

Methanolic solution of each phenothiazine drug(Tables 1-5) (1 mg/ml) was freshly prepared.

Procedure:

a- One to five µl of methanolic solution of each phenomentation drugs were spotted on thin layer plate and allowed to dry. After that 5 µl from each studied amines was overspotted on each applied spot and left to dry. Finally 5 µl NBS was spotted on each previously mixed spot.

After air drying the produced colours were examined in day light.

This method was repeated three times but using instead of NBS, ceric sulphate or chlorine water or potassium periodate.

b- The above procedure was repeated by using the aformentioned reagents as sprayers. This was done by spraying the amine solution upon the developed chromatogram, followed by the oxidant solution. After air drying, the developed colours were observed in day light.

RESULTS AND DISCUSSION

About 50 new reagents for detection of phenothiazine drugs on thin layer plates have been developed (Table 1-4). Each reagent consists of a combination of an amine (aromatic or aliphatic; primary or secondary) and an oxidant. Most of the developed reagents gave various colours with all tested phenothiazine drugs. The sensitivity for detection of these drugs was found to range between 1-5 μ g. The colour developed immediately and remained stable for at least 24 hours.

The type of oxidant plays an important role in the reaction. From 13 oxidants tested (Table 5) only four produced different colours with phenothiazine drugs in presence of the specific amine. The other oxidants gave the same colour or fail to develop any colour with all the examined drugs.

It was found that any variation in the amine or oxidant coupled reagents produced different colours with the same phenothiazine. Moreover; it was observed that the sensitivity of an oxidant in presence of the specific amine decreased in the following order NBS > ceric sulphate > chlorine water > potassium periodate.

Application of these reagents in the form of sprays for detection of phenothiazines on thin layer plates gave favourable results. Some developed reagents were applied successfuly for the detection of phenothiazine drugs in certain pharmaceutical preparations without

prior separation.

Trials have been made to extend the application of the developed reagents for detection of intact phenothiazine drugs in biological fluids. This was done by using combination of morpholine and NBS reagents for detection of promethazine HCl in urine. The produced blue colour is identical with that obtained with reference drug applied on the same plate.

The developed reagents are simple in preparations and handling compared with the more elaborate and corrasive previously reported reagents.

owing to the large number of the developed reagents, and the different colours produced with them when reacted with phenothiazines, it is so difficult to investigate reaction mechanisms which are not clear. However, it is suggested that phenothiazines interact with amines in presence of oxidant to form methylene blue-like dyes.

Further work on the determination of phenothiazines and the nature of products formed in the reaction with certain selected developed reagents is in progress.

zine MH: mepazine HCl, PM: per CPH: Chlorpromazine HCl, LMPM: maleate, TRH: thioridazine HCl maleate, Tainemaleate blue, FPH: green, oxomemazine fluphenazine NC **violet** levomepromazina maleat trifinopromaz olour TPPB thioproper orange יש 対に工作 tartrate dine thiethylperazine BPM: butagera see PC: pericyamorpholine p-nitroaniline red

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Gr.Bl:

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Table 4:	Colour	ĎΩ Hy	TLC spots	FO F	phenothia	zines	with	₫ife	rent	amines	ದ್ದಿ	retassi	E L
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Table 5: Response of the studied phenothiazines to different amine and oxidant reagents.

NO.	Oxidant*	Colours on TLC with all studied phenothia- zine drugs
1.	N-Bromosuccinimide	different colour
2	Ceric sulphate	, ,
3	Chlorine water	5 5
14	Potassium periodate	, ,
5	Potassium permanganate	All brown spots
6	Potassium dichromate	,, ,, yellowish green
7	Iodine solution	,, dark brown
8	Ferric chloride	,, red
9	Potassium ferricyanide	,, grey
LO	Hydrogen peroxide	no colour
L1	Potassium chlorate	, , , ,
12	Potassium bromate	, , , ,
L 3	Potassium iodate	, , , ,

^{*} Each oxidant was tested with all studied amines

REFERENCES

- 1) J. Vecerkova, M. Sulcova and K. Kacl, J.Chromatogr. (Amsterdam) 7, 527 (1962).
- 2) W. Awe, and W.W. Schulze, Pharm. Ztg., 107, 1333(1962).
- 3) A. Woirfalise, J. Chromatog. <u>20</u>, 61 (1965).
- 4) J. Vecerkova, M. Sulcova and K. Kacl, Pharmazie, 17, 22 (1962).
- 5) W. Paulus, W. Hoch and R. Keymer, Arzneimittel-forch., 13, 609 (1963).
- 6) L. Vignoli, B. Cristau, F. Gouezo and J.M. Vassalo, Bull. Trav. Soc. Pharm. Lyon, 9, 277 (1966).
- 7) J. Sochin, and J.W. Daly, J. Pharm. Exp. Therapeut., 139, 160 (1963).
- 8) S.R. El-Shabouri, A.M. Taha and S.A. Hussin, Pharmazie, 37, 71 (1982).
- 9) E.C.G. Clarke " Isolation and Identification of Drugs", Pharmaceutical press, London, (1969) P. 517

استنبط فى هذا البحث حوالى ٥٠ كاشف جديد لادوية الفينوشيازين على رقائق الكروماتوجرافيا وذلك بتكوين ألوان مختلفة وكالما كاشف يتكون من أمين وعامل مو كسد ومعظم هذه الكواشف لها درجال حساسية عالية لائتشاف هيذه الادوية تصل الى ١ مكجم٠

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